

Real World Emissions Analysis Using Sensor-based Emissions Measurement System for Light-duty Direct-Injection Gasoline Vehicle

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Research Background

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✓ Local roadside emission → "Hot spot" ✓ RDE regulations using PEMS





Distance-based NO_x mass emissions for diesel passenger vehicle

PEMS measurement

On-road measurement data; NOx mass ≥ 0.08 g/km

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• Studying the cause of the emission.



- Where, how much and why pollutants are emitted?
- Conduct real-driving experiments on light-duty direct injection gasoline vehicle
- Measure PM/PN, NO and NH₃ emissions by using SEMS (Sensorbased Emission Measurement System)

Test Vehicle and Sensor-based Emission Measurement System



Fuel injection	DI	
Engine type	In-line 4 cylinder gasoline turbo	
Displacement	1,618 cc	
Max. power output	140 / 5600 kW / rpm	
Aftertreatment devise	TWC	
Vehicle mas	1,565 kg	
Emission standard	2005	
Model year	2014	



Emissions Measurement Sensors





NO_x sensor detected NH_3 as well as NO_x

NH₃ Concentration Calculation



Current When NOx sensor and potential sensor for NO are used, ammonia emitted from gasoline vehicles will be measured.

> Sensor signals were compared with those obtained by FT-IR and laser-based measurement system.

[NOx sensor] – [Potential sensor for NOx] $\approx NH_3$

~NOx

Division by 0.9 is used to calibrate the sensor sensitivity.

NH3 sensor for diesel



Not me



Ref: K. Tanaka et al., the 10th Annual International PEMS Conference (2021)

Ref: K. Tanaka et al., Society of Automotive Engineers of Japan, 2020 Annual Autumn Conference Proceedings, No. 232 (in Japanese)

On-road Driving Test Routes





Data sampling rate	10Hz	
Number of	8 times	
measurement		
Fuel	Gasoline	
	(HC1.8)	

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Average Emission



Test Data	Hot/Cold
#1	Cold
#2	Cold
#3	Hot
#4	Cold
#5	Hot
#6	Cold
#7	Hot
#8	Hot





PM/PN Emission Analysis







NO Emission Analysis





NH₃ Emission Analysis





Accumulated NH₃ Mass Emission



 $\begin{array}{c} \text{CO} + \text{H}_2\text{O} \rightarrow \text{CO}_2 + \text{H}_2\\ \text{2NO} + 2\text{CO} + 3\text{H}_2 \rightarrow 2\text{NH}_3 + 2\text{CO}_2\\ \text{2NO} + 5\text{H}_2 \rightarrow 2\text{NH}_3 + \text{H}_2\text{O} \end{array}$

Ref: Barbier et al., Applied Catalysis B Environmental, Vol. 4, p. 105-140, 1994

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PM Emission Hot-spot









NO Emission Hot-spot





NH₃ Emission Hot-spot







Summary



- 1. PM emission increased during rapid acceleration and even more so when vehicle reaccelerates after deceleration.
- 2. NO emission was found especially in lean condition where the excess air ratio is close to 1.
- 3. NH_3 emission increased in rich condition. In rich condition, there is not enough oxygen for complete combustion and CO was formed. NH_3 was generated from this CO.
- 4. NH₃ emission in the first 10 minutes of cold-start was found several times higher than in hot-start and the rest of the data. When the engine is cold, injected fuel is not completely oxidized which leads to the formation of CO and NH₃.
- 5. PM, NO and NH_3 emission tends to occur before and after traffic lights. However, each component hotspot tends to be at different places.
- * For more details, please refer our SAE technical paper, #2022-01-0572



Hypothesis of cold start NH_3 emission due to the existence of CO will be verified by the driving test with SEMS and PEMS.





Thank You for Your Listening

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On Road Driving Test Data	Driving time [s]	Driving Distance [km]	Average Speed [km/h]	Maximum Speed [km/h]	Hot/Cold Condition
#1	1966.8	10.165	18.61	48.14	Cold
#2	1857.9	10.455	20.26	58.90	Cold
#3	2131.3	10.461	17.67	60.09	Hot
#4	1941.0	10.445	19.37	59.56	Cold
#5	2174.1	10.417	17.25	60.08	Hot
#6	2108.6	10.456	17.85	58.09	Cold
#7	2256.6	10.425	16.63	56.12	Hot
#8	2100.4	10.445	17.90	58.11	Hot