

Emission Measurement with NOx Sensors in the Presence of HC's & H<sub>2</sub>

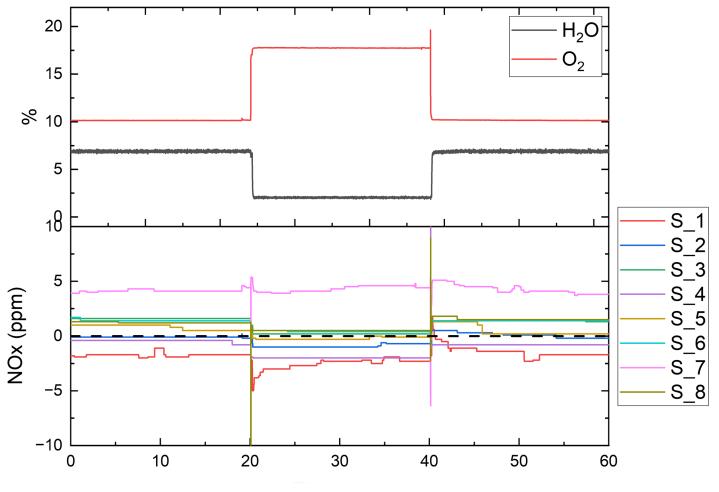
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## *State* **1**: 7% H<sub>2</sub>O, 10% O<sub>2</sub>, 6.5% CO<sub>2</sub> *State* **2**: 2% H<sub>2</sub>O, 17.5% O<sub>2</sub>, 2% CO<sub>2</sub>

Oppm NOx, fast O<sub>2</sub> Change: positive drift @state 1 drift lower @ state 2



Time (min)

#### Background

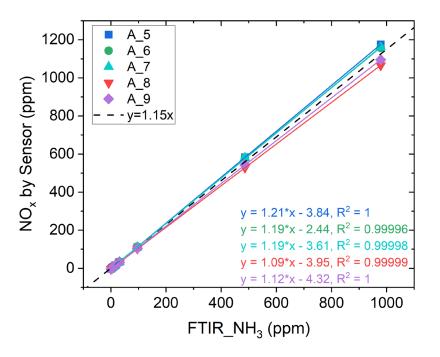
- On-board monitoring relies on the vehicle's sensors to continuously monitor emissions for emissions compliance, system repair, or fleet emissions inventory measurement.
- Understanding the performance of NO<sub>x</sub> sensor is crucial for both lean and stoichiometric internal combustion engine (ICE) applications as their sensitivities may vary.
- NO<sub>x</sub> sensors measure both O<sub>2</sub> and NO<sub>x</sub> while being cross-sensitive to several species on each channel.
- Here we characterize and discuss NO<sub>x</sub> sensor cross-sensitivities to various reducing agents in the context of stoichiometric CNG engines and H<sub>2</sub>-ICE applications.

# Potential NO<sub>x</sub>/NH<sub>3</sub> Emission Deconvolution Method for Stoichiometric Applications

- NO<sub>x</sub> sensors have a well known cross-sensitivity on their NO<sub>x</sub> channel to NH<sub>3</sub> (see right).
- For stoichiometric-ICE applications both NO<sub>x</sub> (lean) and NH<sub>3</sub> (rich) may be present.
- The proposed method relies on the NO<sub>x</sub> sensor's O<sub>2</sub> channel to arbitrate between when the NO<sub>x</sub> channel is responding to NO<sub>x</sub> (lean) vs NH<sub>3</sub> (rich).

#### **Outstanding Questions:**

- 1. What species/factors impact the sensor's O<sub>2</sub> readings and may negatively impact the sensor's ability to arbitrate between rich and lean vehicle operation?
- 2. Is the sensor's  $NH_3$  cross-sensitivity consistent across both rich and lean operation?
- 3. What are the implications of the sensor's  $H_2$  sensitivity for lean burn H2-ICE applications?
- 4. Are there any situations on stoichiometric systems, healthy or failed, in which you will have the simultaneous presence of  $NO_x$  and  $NH_3$  or  $NO_x$  when it is rich and  $NH_3$  when it is lean?
- 5. Are OEM and aftermarket NO<sub>x</sub> sensors capable?



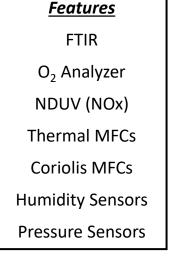
#### Methods: Sensor Exerciser and Test Instrument (SETI)

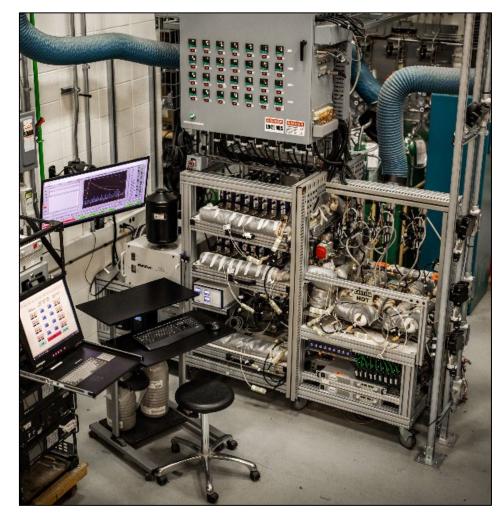
Primary sensor characterization test bench:

- Temperature Range: 20-550 °C.
- Isobaric flow sweep capability (<40 psi).
- Flow Range: 20 250 SLPM.
- ≈0.2 sec response time measurements.
- Independently controlled H<sub>2</sub>, CH<sub>4</sub>, C<sub>3</sub>H<sub>8</sub>, CO, CO<sub>2</sub>, N<sub>2</sub>, O<sub>2</sub>, H<sub>2</sub>O, NO, NO<sub>2</sub>, & NH<sub>3</sub> concentrations.

Here we tested with two HC mixtures:

- Mix1: 1% CO, 3000 ppm H<sub>2</sub>, 3000 ppm CH<sub>4</sub>, 0 ppm NO, at 150 °C.
- Mix2: 0% CO, 0 ppm H<sub>2</sub>, 2200 ppm CH<sub>4</sub>, 0 ppm NO, at 350 °C.





### **Concept Introduction: "Net O<sub>2</sub>"**

- Net O<sub>2</sub> is the O<sub>2</sub> concentration less the concentration of reducing agents as considered based on the amount of O<sub>2</sub> needed to fully oxidize said reducing agents.
- OBD-style NO<sub>x</sub> and O<sub>2</sub> sensors are designed/expected to measure "Net O<sub>2</sub>" and not O<sub>2</sub> concentration.
- When "Net O<sub>2</sub>" = 0% we consider this to be the definition of the λ=1, stoichiometric condition. All "rich" conditions are defined by Net O<sub>2</sub> concentrations less than zero whereas "lean" conditions correspond to positive Net O<sub>2</sub> concentrations.
- Example Calculation:
  - O<sub>2</sub> concentration = 10% as measured by a reference analyzer (e.g. paramagnetic)
  - $CH_4$  concentration = 2% as measured by a reference analyzer (e.g. FTIR)
  - H<sub>2</sub> concentration = 1% as measured by a reference analyzer (e.g. mass flow meters)

$$CH_4 + 2 \cdot O_2 \rightarrow CO_2 + 2 \cdot H_2O \tag{1}$$

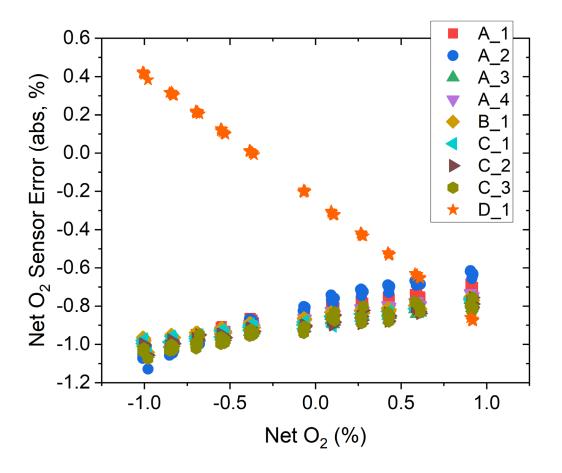
$$H_2 + \frac{1}{2} \cdot 0_2 \to H_2 0$$
 (2)

• Net  $O_2 = 10\% - 4\%$  (CH<sub>4</sub>) - 0.5% (H<sub>2</sub>) = 5.5%

# $O_2$ Channel Accuracy with $NH_3$ vs. $\lambda$ (HC Mix 1)

Test condition:

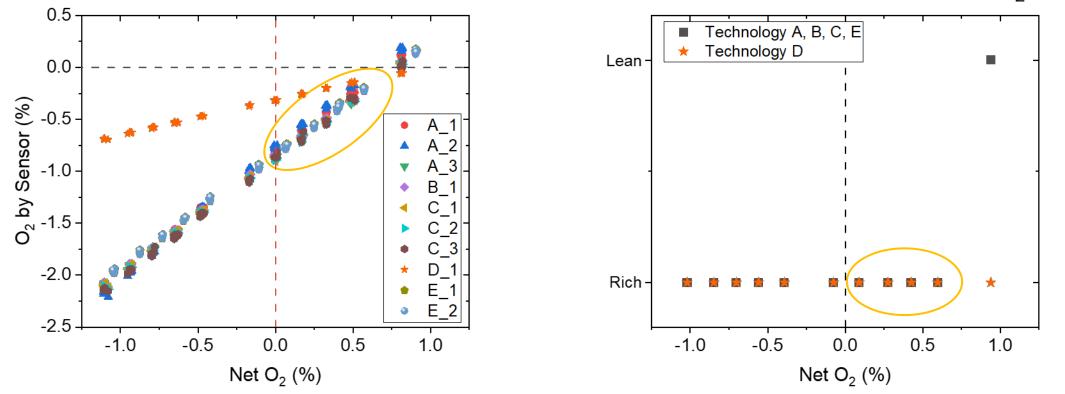
- 1% CO, 3000ppm H<sub>2</sub>, 3000ppm CH<sub>4</sub>, 0ppm NO
- 150°C
- NH<sub>3</sub>: 256, 128, 64, 32, 16, 8, 0ppm (multiple points for each sensor at each O<sub>2</sub> concentration)



- Sensor technologies A, B, C are showing a weak positive dependence as O<sub>2</sub> increases and an overall negative bias.
- Sensor technology D is showing a negative dependence as O<sub>2</sub> increases.

Prepared by Cummins for OSAR

# O<sub>2</sub> Arbitration – Lean vs Rich with HC Mix 1?

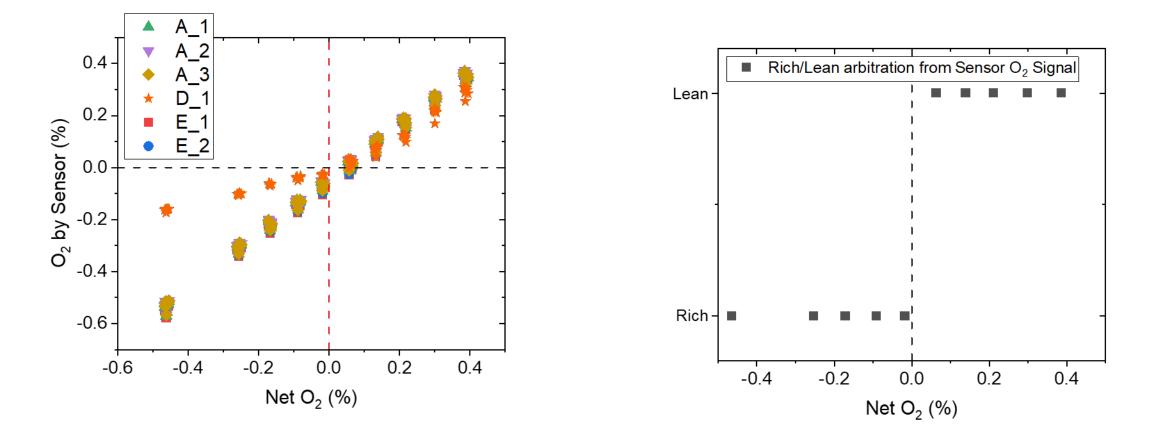


Rich/Lean arbitration from Sensor O<sub>2</sub> Signal

- 1% CO, 3000 ppm H<sub>2</sub>, 3000 ppm CH<sub>4</sub>, 0 ppm NO, at 150 °C.
- The NO<sub>x</sub> sensors are under-reporting the O<sub>2</sub> level and read rich when it's lean.

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# O<sub>2</sub> Arbitration – Lean vs Rich with HC *Mix 2*?

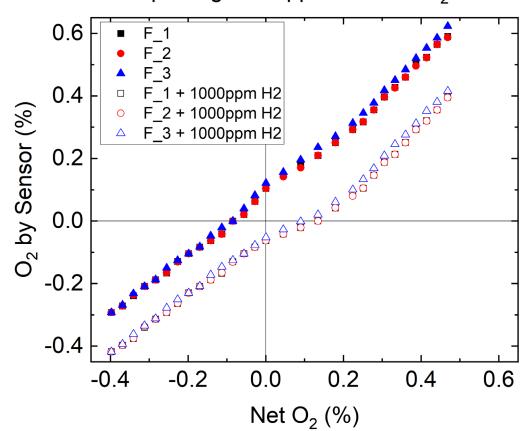


- 0% CO, 0 ppm H<sub>2</sub>, 2200 ppm CH<sub>4</sub>, 0 ppm NO, at 350 °C.
- NH<sub>3</sub>: 256, 128, 64, 32, 16, 8, 0ppm.
- The NO<sub>x</sub> sensors are *slightly* under-reporting the O<sub>2</sub> level and could distinguish rich vs lean.

#### **Sensor Supplier Observations:**

HC Mixtures of Propane, CO, and H<sub>2</sub>

- Schaeffler tested with fixed concentrations of C<sub>3</sub>H<sub>6</sub> and variable amounts of O<sub>2</sub>, CO, & H<sub>2</sub>.
- Two tests were done with one test exposing the sensors to 1000 ppm more  $H_2$  at each point which was offset by a 1000 ppm decrease in CO (unchanged  $\lambda$ ).
- Sensors exposed to higher H<sub>2</sub> concentration were biased lowe by the exposure.
- Presence of the heavy molecule C<sub>3</sub>H<sub>6</sub> might be the cause of the observed lean bias under some conditions.

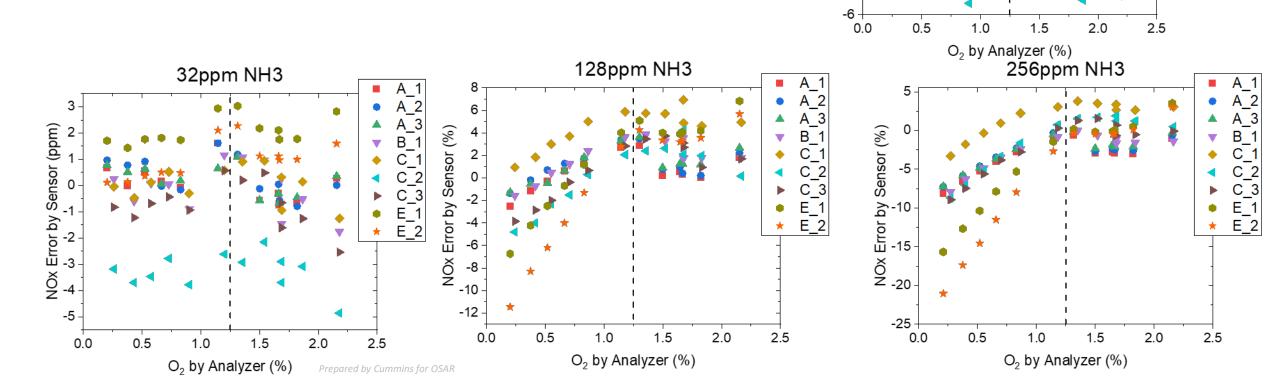


Replacing 1000 ppm CO with H<sub>2</sub> 1:1

Data provided courtesy D. Wieland, Schaeffler Automotive Group

### NO<sub>x</sub> Channel NH<sub>3</sub> Sensitivity Variability vs O<sub>2</sub>

- 1% CO, 3000ppm H<sub>2</sub>, 3000ppm CH<sub>4</sub>, 0ppm NO, 5% CO<sub>2</sub>, 20% H<sub>2</sub>O at 150C.
- The NO<sub>x</sub> sensor's NH<sub>3</sub> cross-sensitivity drops off as Net O<sub>2</sub> crosses zero.
- Here errors are computed after pressure compensating and by utilizing the supplier published NH<sub>3</sub> cross-sensitivity factors.



0ppm NH3

NOX Error by Sensor (ppm)

A\_1

В

E 2

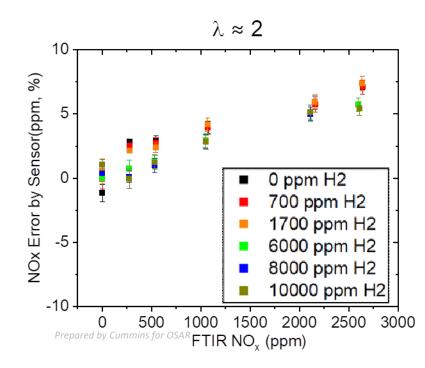
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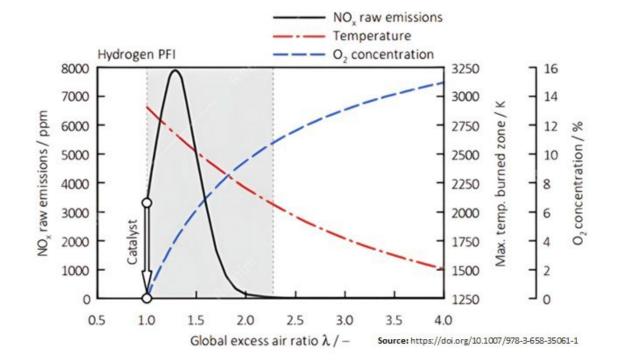
# **Applicability for H<sub>2</sub>-ICE**

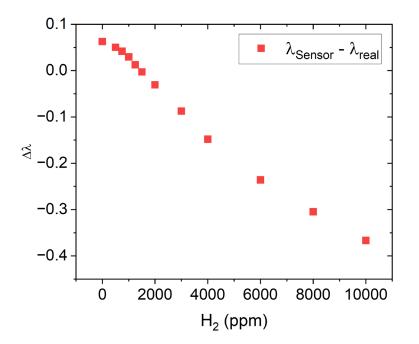
• The measured  $H_2$  sensitivity on the  $O_2$  channel is ~4x the sensitivity needed for the sensor to serve as a consistent indicator of  $\lambda$  (i.e. ~2 vs ½).

$$\mathrm{H}_{2} + \frac{1}{2} \cdot \mathrm{O}_{2} \rightarrow \mathrm{H}_{2}\mathrm{O}$$

• The  $NO_x$  channel is largely insensitive to  $H_2$ .







#### **Conclusions & Next Steps:**

Conclusions:

- We found that the sensor's O<sub>2</sub> readings are particularly impacted by H<sub>2</sub> which induces a rich bias under both truly rich and truly lean conditions alike.
- If H<sub>2</sub> were to exist under lean conditions, it could lead to a sensor's broadcast NOx values to be erroneously associated with NH<sub>3</sub> emissions.
- Both positive and negative biases of the sensor's O<sub>2</sub> channel are possible with highly diffusive reducing agents biasing the net O<sub>2</sub> reading low and slower molecules potentially imposing a high bias.
- NH<sub>3</sub> sensitivity remains *relatively* consistent around  $\lambda$ =1 especially at lower NH<sub>3</sub> levels.
- The method considered for NO<sub>x</sub> monitoring on stoichiometric applications may have issues under some conditions while working fine under others.

#### **Potential Next Steps:**

- 1. Get a better understanding of the real-world H<sub>2</sub> + HC + CO concentrations at the tailpipe for both healthy and degraded systems.
- 2. Consider a combined  $NO_x+NH_3$  threshold based approach.
- 3. Characterize alternatives technologies:
  - Aftermarket NOx sensors
  - Traditional wideband and narrowband O<sub>2</sub> sensors
  - Prepare H2 sensing technologies for compensation



