# Issues and Research at UCR on PM and Toxics Impacts of VOCs

William P. L. Carter CE-CERT, University of California, Riverside, CA May 28, 2009

#### Outline

- Factors affecting impacts of VOCs on secondary toxics and modeling study of aldehyde and PAN formation potentials of VOCs
- Factors affecting impacts of VOCs on PM and qualitative UCR chamber data on relative impacts of coatings VOCs
- Summary of issues and research needs for developing reactivity scales for impacts of VOCs on SOA and toxic products

# Factors Affecting Impacts of VOCs on Air Toxics Formation

#### **Direct Effects**

- For toxic VOCs the main factors are their removal rates
- For toxic precursor VOCs the main factors are the rate the VOCs react, the toxic product yields and how they vary with conditions

#### **Indirect Effects**

- VOCs that enhance or inhibit radical levels will affect rate of toxic product formation from toxic precursors that are present
- VOCs that affect O<sub>3</sub> will affect the chemical environment that may affect some toxic product yields (e.g., PANs)
- Indirect effects will be highly dependent on the levels of toxic precursors present
- Indirect effects will correlate with O<sub>3</sub> reactivity more than direct effects.

# UCR Project for OEHHA to Estimate Aldehyde and PAN Formation Potentials

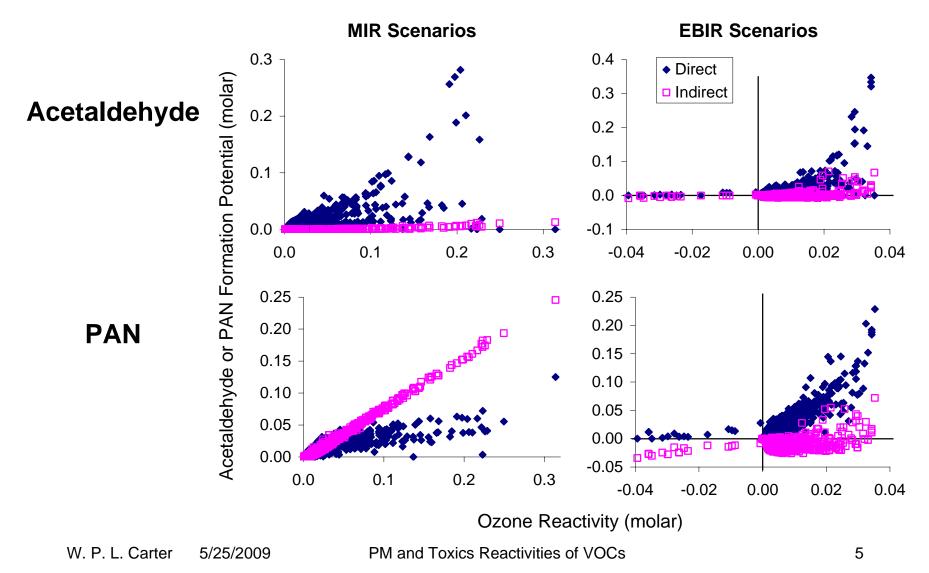
#### Background

- The California Office of Environmental Health Hazard Assessment needed to evaluate the human and environmental health impacts of motor vehicle fuels
- Quantifying effects on secondary toxics of fuel VOCs on air toxics (aldehydes, nitrates, PANs) was part of this effort
- UCR was contracted to calculate impacts of VOCs on atmospheric aldehyde, nitrate, and PAN levels
- Work was completed in September, 2001. Report now available at http://www.cert.ucr.edu/~carter/absts.htm#aldrpt

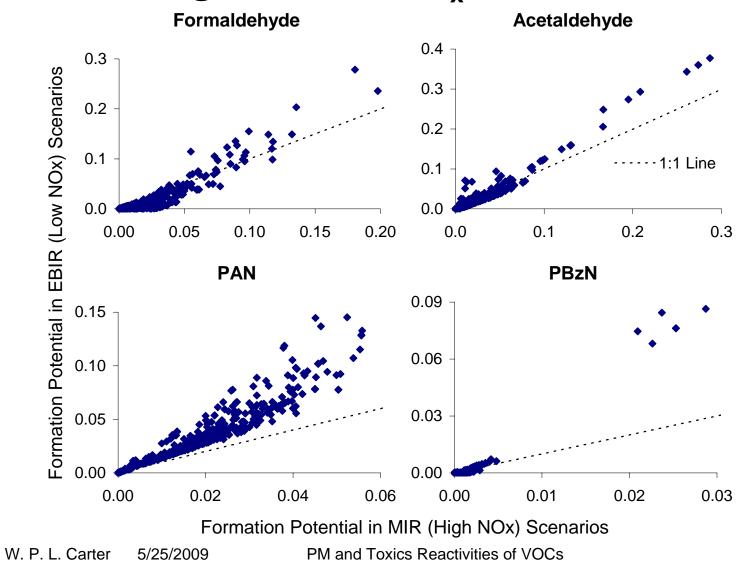
# Project to Estimate Aldehyde and PAN Formation Potentials: Approach

- SAPRC-99 mechanism used
- Duplicate organic product species added to permit separate assessments of direct and indirect effects
- Same scenarios employed as used to calculate MIR, MOIR, and EBIR ozone reactivity scales
- Incremental reactivities of VOCs were calculated with respect to direct and indirect formation of:
  - Formaldehyde PAN
  - Acetaldehyde PBzN
  - Lumped higher
    Lumped Other
    Aldehydes
    PANs
- Acrolein
- Organic nitrates
  - Lumped aromatic products

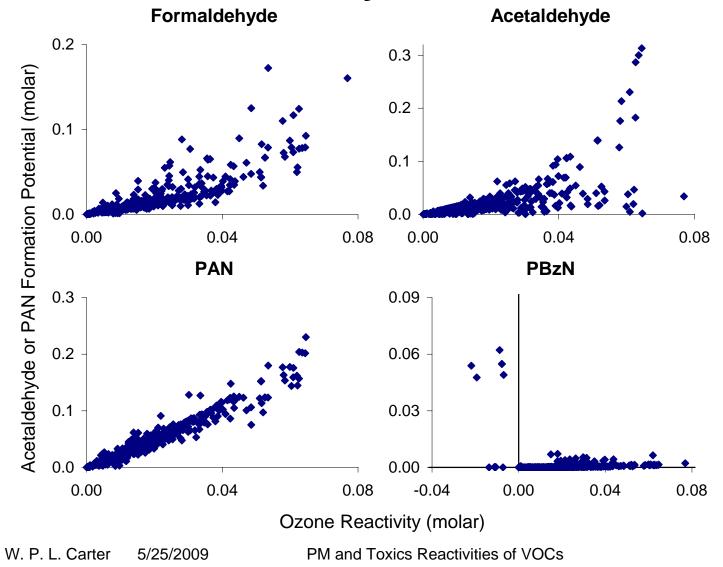
#### Direct and Indirect Acetaldehyde and PAN Formation Potentials vs. Ozone Reactivity



#### Aldehyde and PAN Formation Potentials in High vs. Low NO<sub>x</sub> Scenarios



#### Aldehyde and PAN Formation Potentials vs. Ozone Reactivity in MOIR Scenarios



# Summary and Recommendations - Toxic Product Reactivity

#### Summary

- Toxic product reactivities depend on details of VOC mechanisms which vary from compound to compound
- Aldehyde formation potentials are insensitive to NO<sub>x</sub> conditions.
  PAN potentials more sensitive but trends are consistent.
- Correlation with O<sub>3</sub> reactivity depends on product. Ranking is PAN > Formaldehyde > Acetaldehyde >> PBzN

#### Recommendations

- Issues and research needs similar to those for  $O_3$  scales except:
  - Greater requirement for chemical detail in mechanism and greater mechanism uncertainty
  - Choice of scenario conditions may be slightly less important except for composition of base ROG mixture.

## Factors Affecting Impacts of VOCs on Secondary PM

- Many VOCs form low volatility oxidation products that can partition into the aerosol phase and contribute to secondary PM
- Some higher volatility products may also partition into the aerosol phase due to heterogeneous reactions
- The yields of condensable products varies from compound to compound and with atmospheric conditions
- Identity, yields, formation mechanisms, partitioning coefficients, and heterogeneous reactions of condensable products are mostly unknown for most VOCs
- Data and mechanistic knowledge are inadequate for models to predict secondary PM from VOCs with any degree of reliability.
- Current models use inadequately tested and highly simplified parameterized models to predict secondary PM

#### **Requirements for Predictions of PM Impacts** (in addition to requirements for O<sub>3</sub> predictions)

#### **Predictive Mechanisms for Chemical and Physical Processes**

- Chemical mechanism must predict yields of low-volatility products and how they change with chemical conditions
- Predict gas-to-particle phase partitioning and evaporation
- Predict condensed-phase reactions (polymerization, etc)
- Predict PM nucleation and removal processes

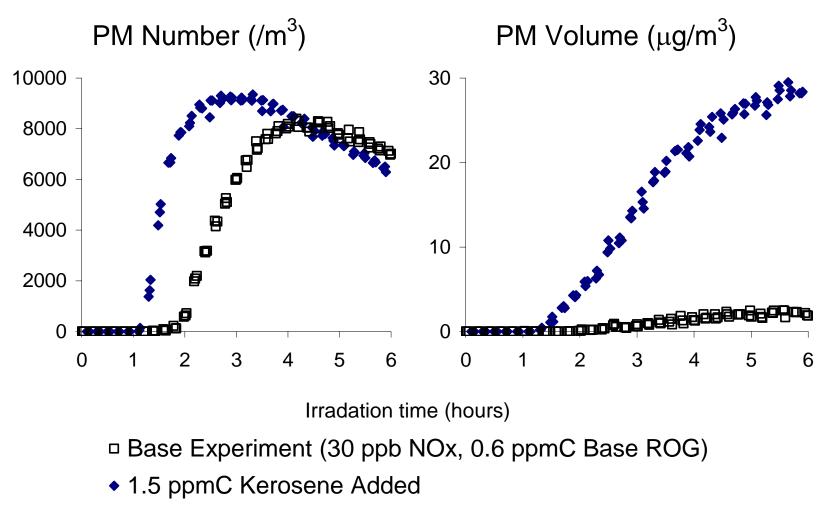
#### Model for Environmental Conditions

- Appropriately representing temperature, humidity, absolute concentrations *much more important* for PM than for  $O_3$
- Appropriately represent *primary* and *background* PM that may affect partitioning or condensed-phase processes
- Use base case scenarios that give representative PM levels
- Appropriately represent PM loss processes

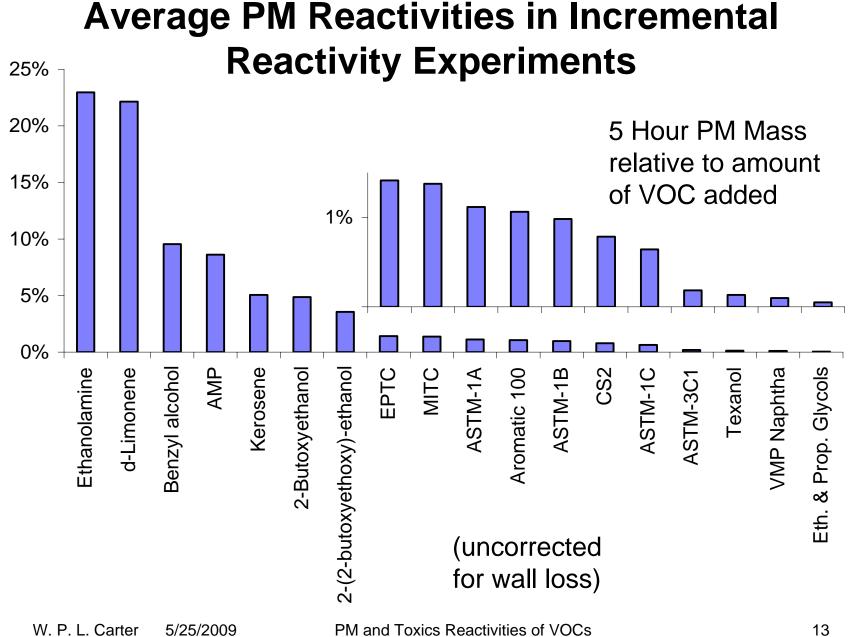
## **PM Measurements in the UCR EPA Chamber**

- PM Measurements were made in conjunction with most recent ozone reactivity experiments. Compounds studied include:
  - Representative Water- and Solvent-based coatings VOCs
  - Several pesticide VOCs: MITC, EPTC, CS<sub>2</sub>, Kerosene
  - Aminomethylpropanol (AMP), Aminoethanol, d-Limonene
- Research underway to determine effects of reactant (e.g. NO<sub>x</sub>) concentrations, humidity, and other factors on PM formation
  - Current emphasis is on aromatics and terpenes
  - Data obtained are needed to develop predictive mechanisms for PM formation in the atmosphere
- Experiments are being carried out at lower reactant concentrations than is practical for most other chambers
- A large array of state-of-science equipment recently was obtained from a grant from the Keck foundation

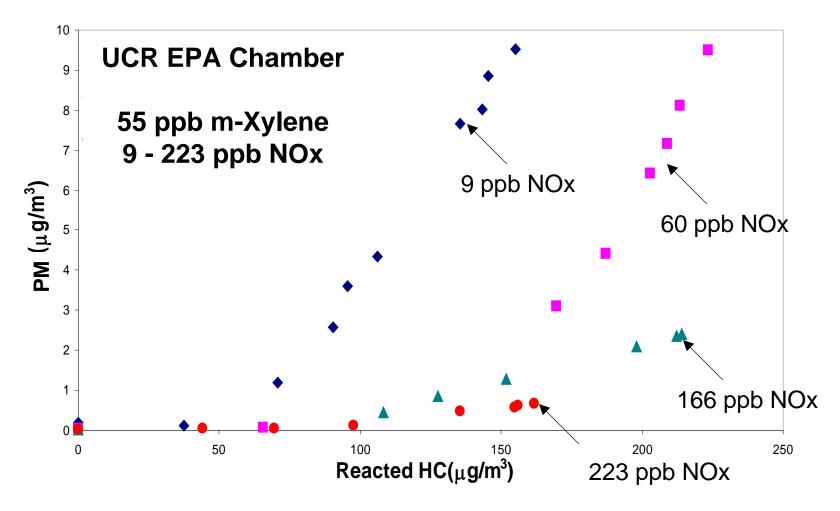
#### **Representative PM Data** from Reactivity Experiments



PM and Toxics Reactivities of VOCs



#### Comparison of SOA in Experiments with Same Initial m-Xylene Concentrations



# Upcoming UCR Project: SOA Formation: Chamber Study and Model Development

- <u>Objectives</u>: Develop improved mechanisms to predict secondary organic aerosol (SOA) from aromatics and other compounds.
- <u>Status</u>: Contract pending
- <u>Proposed Overall Approach</u>:
  - This project to be carried out in collaboration with Dr. David Cocker at UCR and Dr. Robert Griffin at Rice University.
  - UCR EPA chamber experiments will be conducted to develop and evaluate models for SOA from aromatics, aromatic products, and selected other VOCs
  - The SAPRC-07 mechanism will be expanded and adapted for SOA modeling based on the results of the experiments.
  - Use the results to develop improved mechanisms for predicting SOA in 3-D airshed models.

# Summary and Recommendations - PM Reactivity: Scientific Issues

- Chamber data indicate VOCs vary significantly in PM impacts, and PM impacts are *not* correlated with O<sub>3</sub> impacts.
  - However, current data are qualitative and not representative of ambient conditions
- Improved, and more comprehensive mechanisms need to be developed for modeling secondary PM in the atmosphere
  - Detailed mechanisms have been developed for only few of VOCs and evaluated under unrepresentative conditions
  - Current simplified PM models are not consistent with the limited available chamber data for aromatics
  - Well-characterized chamber experiments *simulating ambient conditions* are needed to develop *predictive* PM models.
- Scenarios for assessing PM impacts need to be developed. Existing scenarios for  $O_3$  reactivity scales are not appropriate

# Summary and Recommendations - PM Reactivity: Policy Issues

- Current models are not yet sufficiently reliable to serve as a basis for deriving PM reactivity scales for regulatory applications
- Near term solution:
  - Use PM formation in appropriate environmental chamber experiments to obtain *qualitative* PM reactivity rankings
  - Modeling and experimental research is needed to develop appropriate experimental procedures for this purpose
- Longer term solution:
  - Continue to support experimental and model development research to improve SOA models.
  - Once sufficiently reliable models are developed, the issues become similar to those for  $O_3$  and toxics reactivity scales.

#### **Additional Information Available**

W.P.L. Carter research on chemical mechanisms and reactivity

• http://www.cert.ucr.edu/~carter

Report on Aldehyde and PAN formation potentials for the California OEHHA (Dated September 23, 2001)

• http://www.cert.ucr.edu/~carter/absts.htm#aldrpt

Research on Secondary Organic Aerosol Formation in UCR EPA chamber

• http://www.engr.ucr.edu/~dcocker/