

Chamber Evaluation of Photochemical Indicators of Ozone Sensitivity to VOC and NO_x

Gail Tonnesen, Chao Jung-Chien, Bill Carter

University of California, Riverside
Bourns College of Engineering
Center for Environmental Research and Technology

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Indicators of O₃ Sensitivity

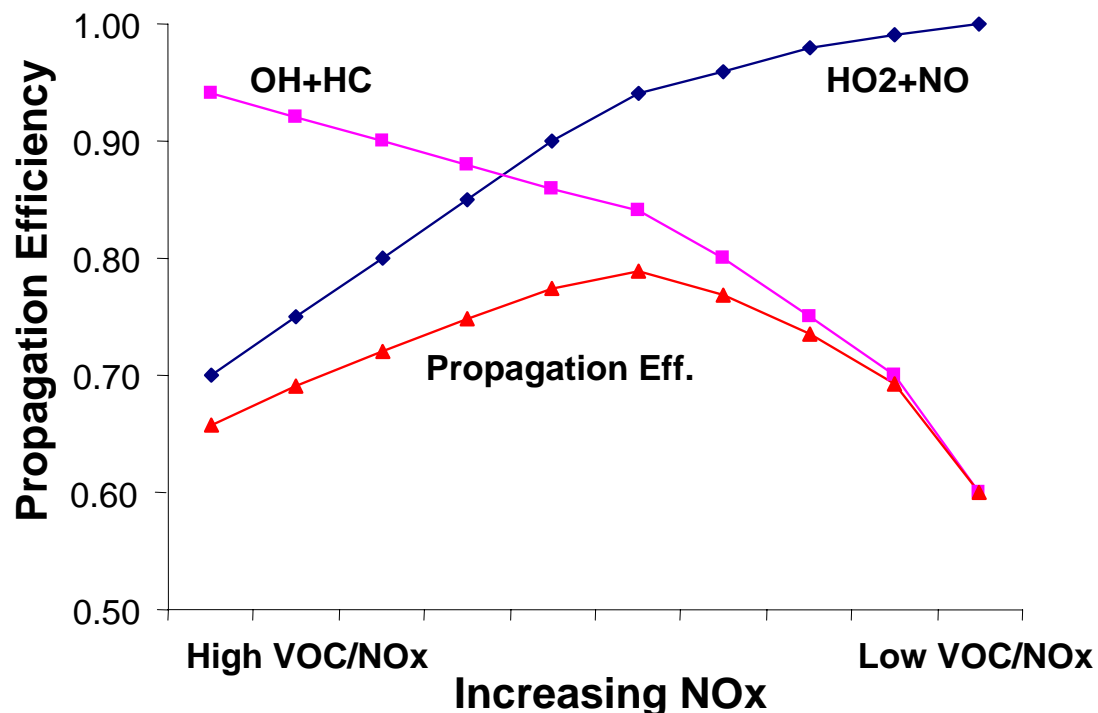
- Desire species or ratios that consistently assume different values under conditions of NO_x-sensitive and VOC-sensitive ozone.
- Theoretical derivations include:
 - Johnson (1984): max potential O₃ for a given NO_x level.
 - Milford et al., (1994): total NO_y
 - Sillman (1995), Kleinman (1994): steady-state radical budgets.
 - Cardelino & Chameides (1994), Kleinman: constrained SS model
 - Tonnesen and Dennis (1997): radical propagation efficiency.

OH Propagation Efficiency

- Derivation based on analysis of radical propagation efficiency.

$$OH: f_{OH+HC} + f_{OH+NO_2} + f_{OH+misc} = 1$$

$$HO_2: f_{HO_2+NO} + f_{HO_2+RO_2} + f_{HO_2+misc} = 1$$

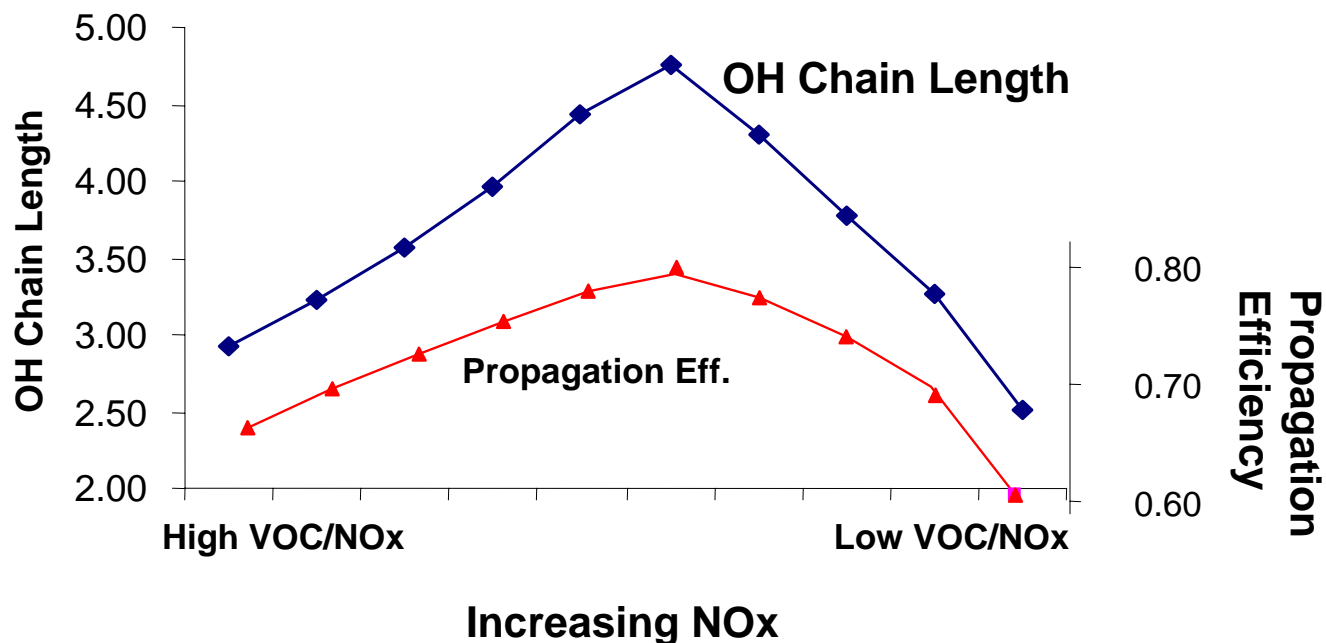


Propagation Efficiency \approx
 $(f_{OH+HC})(f_{HO_2+NO})$

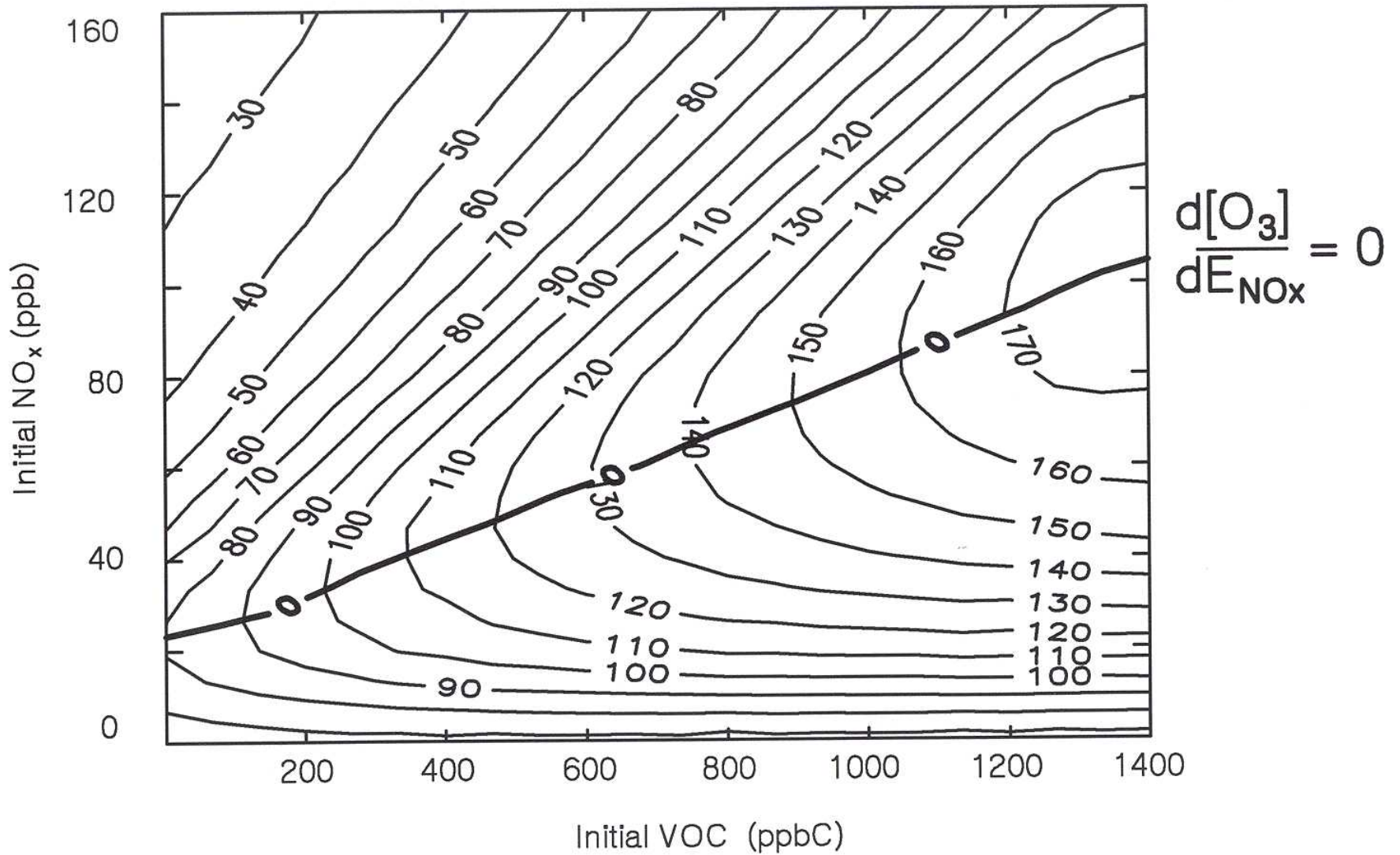
OH Chain Length

$$\text{Chain Length} = 1 + Pr + Pr^2 + Pr^3 + \dots = 1 / (1 - Pr)$$

Chain length increases rapidly with small increase in Prop



Peak [O₃] (ppb) RADM2 Mec, base case



Indicators of O₃ Sensitivity

- Unique values of each indicator are associated with the ridgeline for both peak [O₃] and P(O_x):

$$\partial [\text{O}_3]/\partial E_{\text{NO}_x} = 0$$

$$\partial P(\text{O}_x)/\partial E_{\text{NO}_x} = 0$$

- No *a priori* estimate of indicator ratio – transition values determined numerically using models.

Indicators of [O₃] Sensitivity



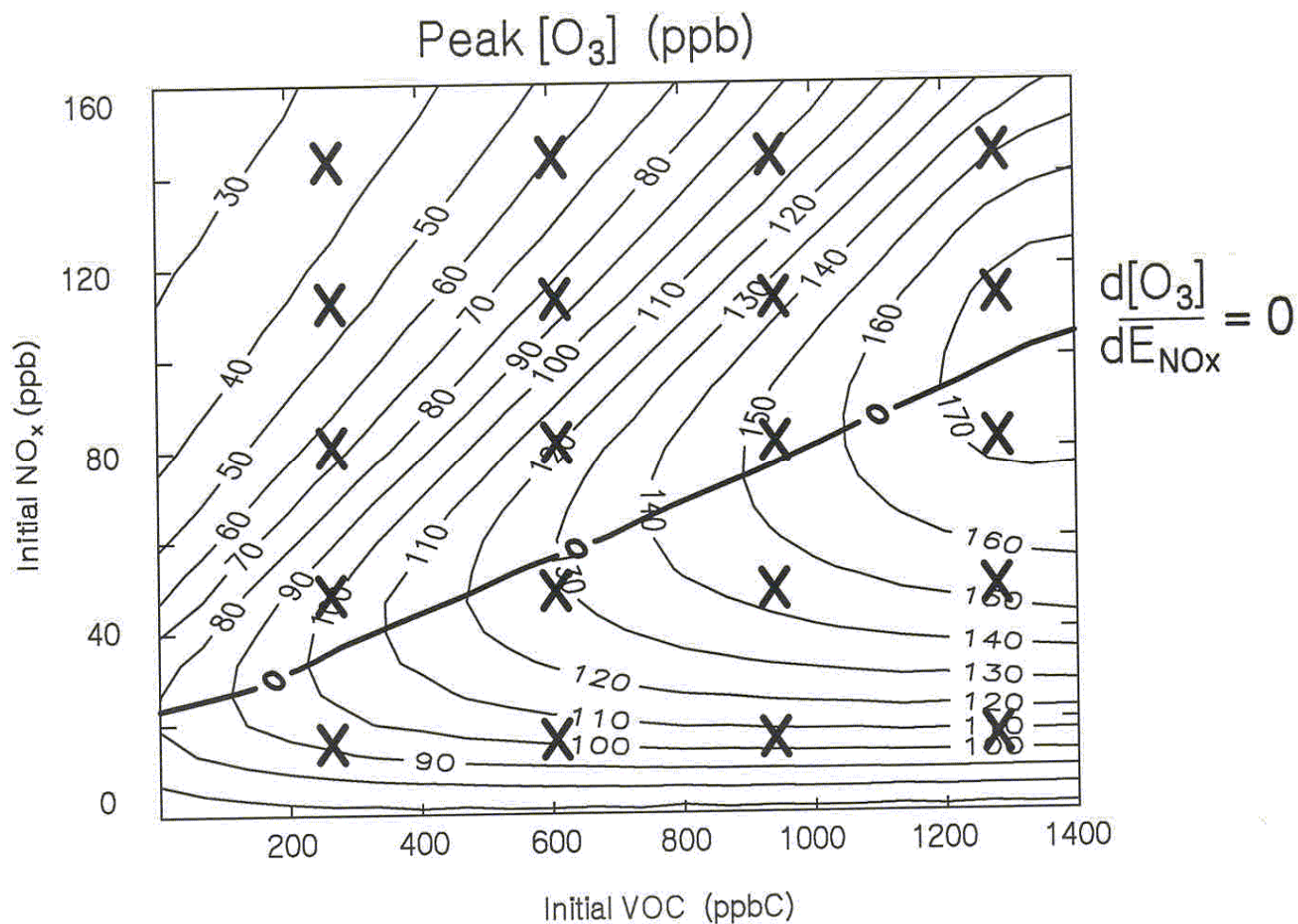
Motivation for this Study

- Strong Theoretical Basis for Use.
 - But based on models with large uncertainties.
- Best method for assessing O₃ sensitivity
 - But models sensitivity experiments show mixed results.
 - Question remain about robustness and discriminating power.
- Experimental validation of the indicator concept is needed using chamber experiments to evaluate indicators in a controlled setting.

Chamber Empirical Evaluation

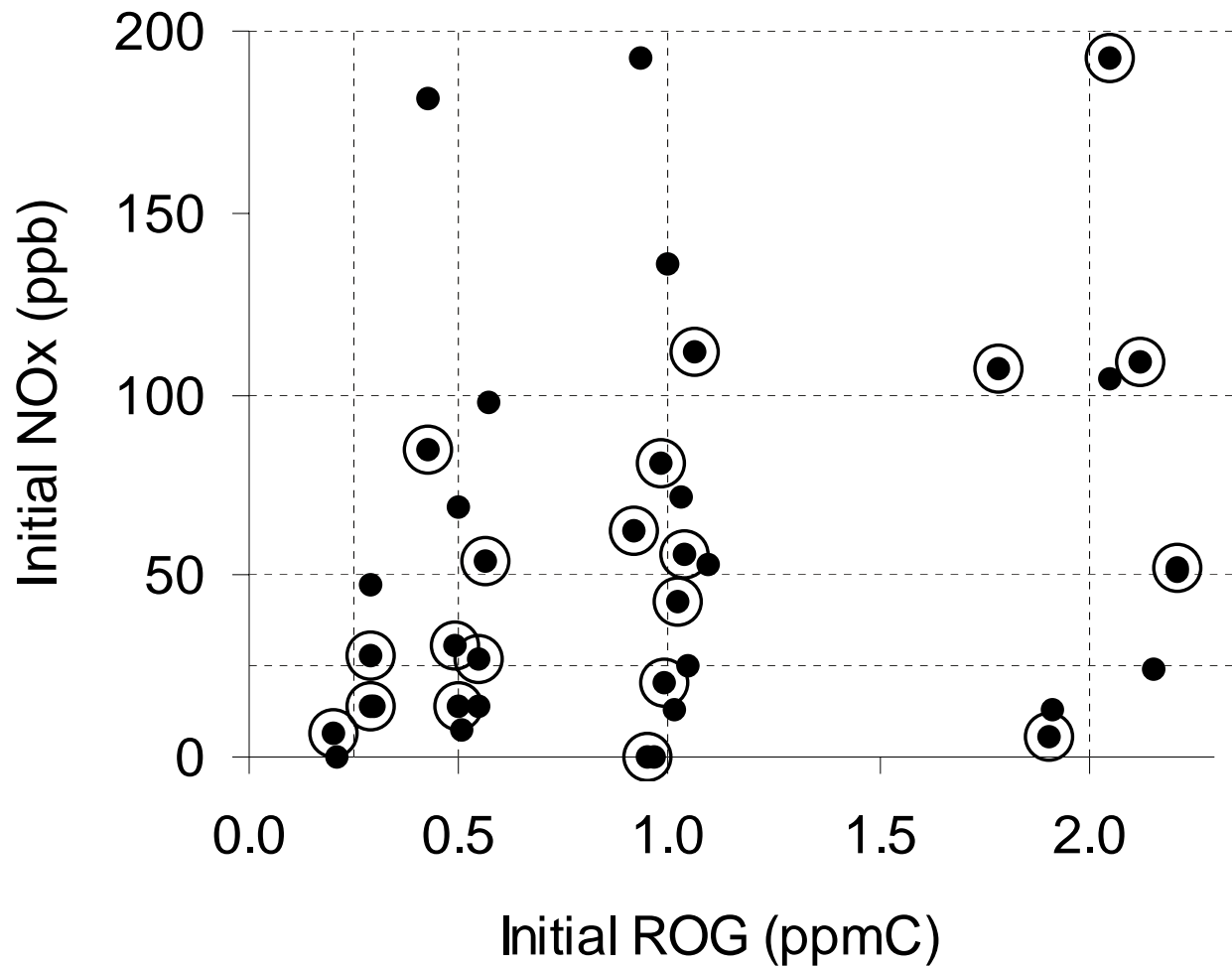
- Evaluation of indicators in chamber experiments:
 - Measure the values of each indicator for a range of VOC and NO_x levels, and determine if a unique value of each indicator is associated with the O₃ ridgeline.
 - Existing chamber data base lacks the measurement needed to quantify HO_x and NO_y budgets. (Need a low NO_x chamber).
 - wall effects still a concern, but these are reduced in the UCR's EPA chamber, and are included in the chamber wall mechanism.

Goal: experiments over a matrix of VOC/NO_x pairs across the O₃ Isopleth Surface



Actual Experimental Matrix

● All Runs ○ Runs with Radical Data



Analytical Instrumentation

Ozone, CO, NO, NO _y , Humidity, Temperature, Total Hydrocarbons	Commercial Ambient Monitoring Instruments
Hydrocarbons	2 GC-FID, GC-ECD
NO (20 ppt MDL)	ECO Physics Chemiluminescence Analyzer
NO ₂ and PAN	2 Luminol-GC Instruments
NO₂ (<0.2 ppb MDL)	2 Dual Channel
HNO₃ (1–2 ppb MDL)	Mid IR TDLAS Systems
HCHO (0.5–1 ppb MDL)	-
H₂O₂ (1–2 ppb MDL)	
Light Spectrum	Spectroradiometer
Light Intensity	Spherical Irradiance Sensor Quartz Tube J(NO ₂) Monitor

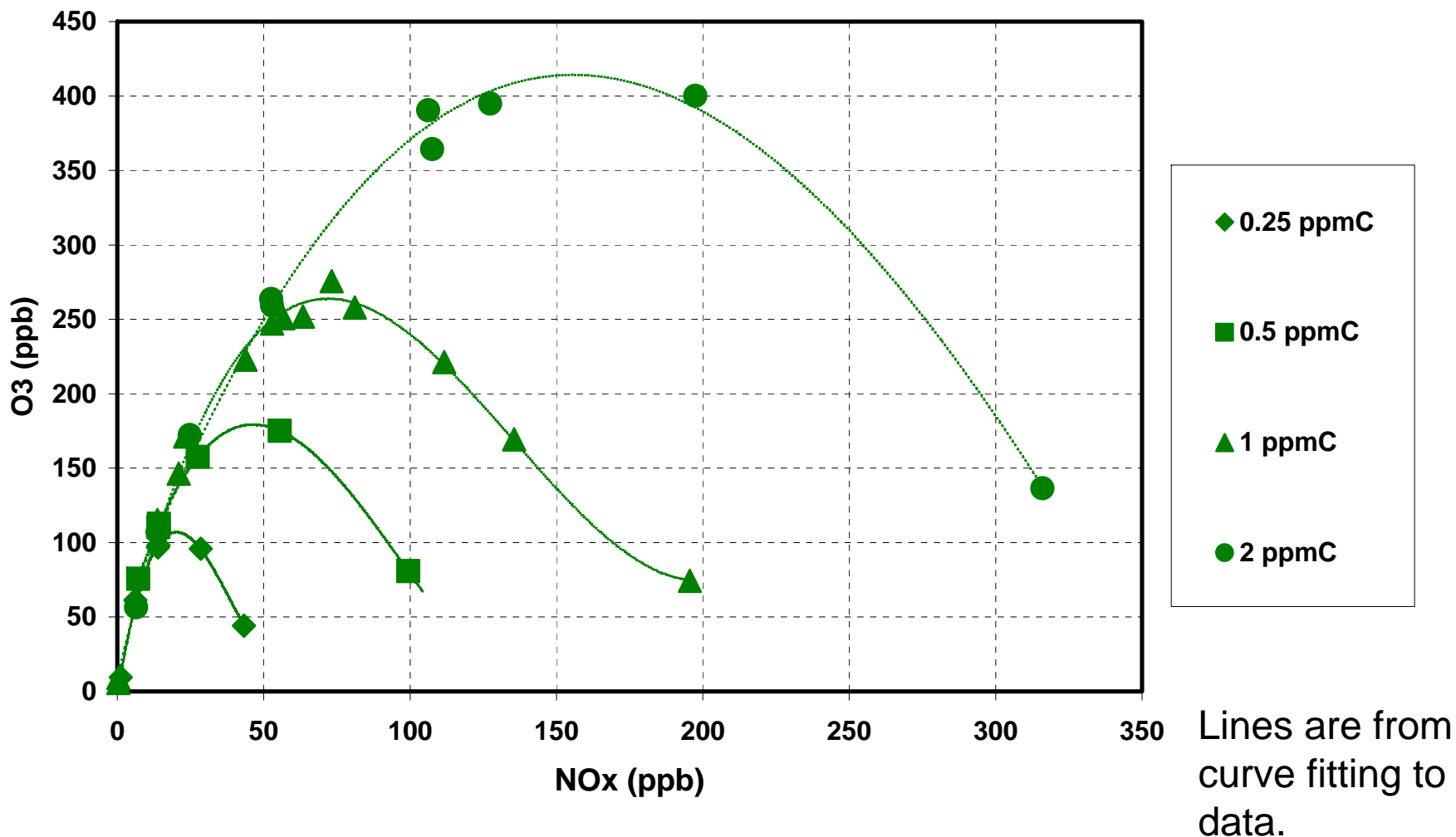
Analytical Instrumentation (2)

- Bill Brune's (PSU) Laser Induced Fluorescence instrument:
 - Measures OH, HO₂ and direct measure of VOC reactivity.
 - LIF detector was located in side A of the Chamber so HO_x data is available for 50% on experiments.
 - Detection limits
 - OH: 0.01 pptv
 - HO₂: 0.1 pptv

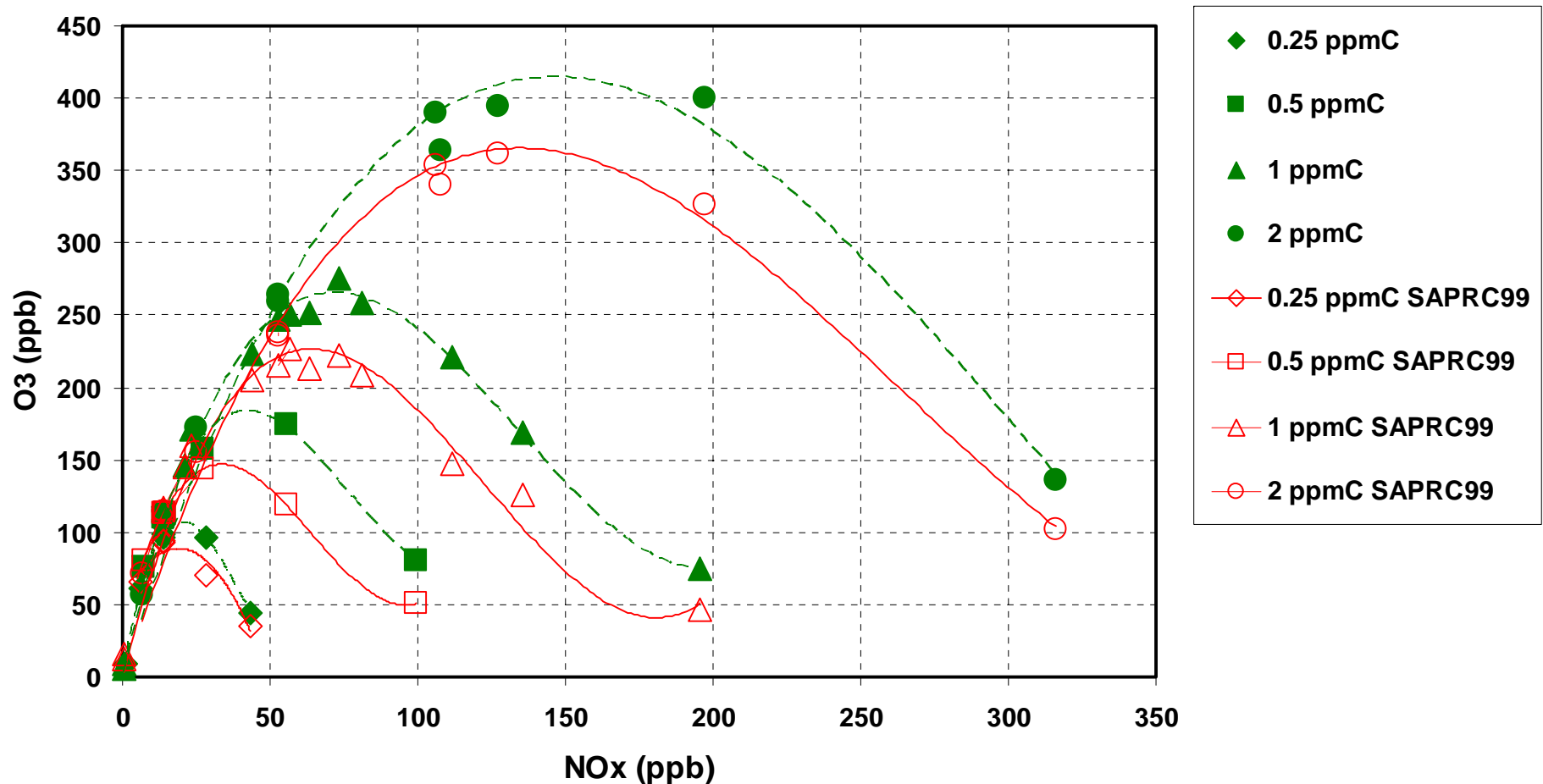
Experimental Plan

- Measure direct reactivity as $P(O_x)$ and compare with indicators of $P(O_x)$ sensitivity to VOC and NO_x .
- Measure O_3 concentration and compare to indicators of peak O_3 sensitivity to VOC and NO_x :
 - Evaluate results at the end of 6 hours.
- Perform mass budget analysis by comparing integral of reactions rates to changes in species conc.
- Compare with model simulations.

Observed O3 conc. at 6th hr

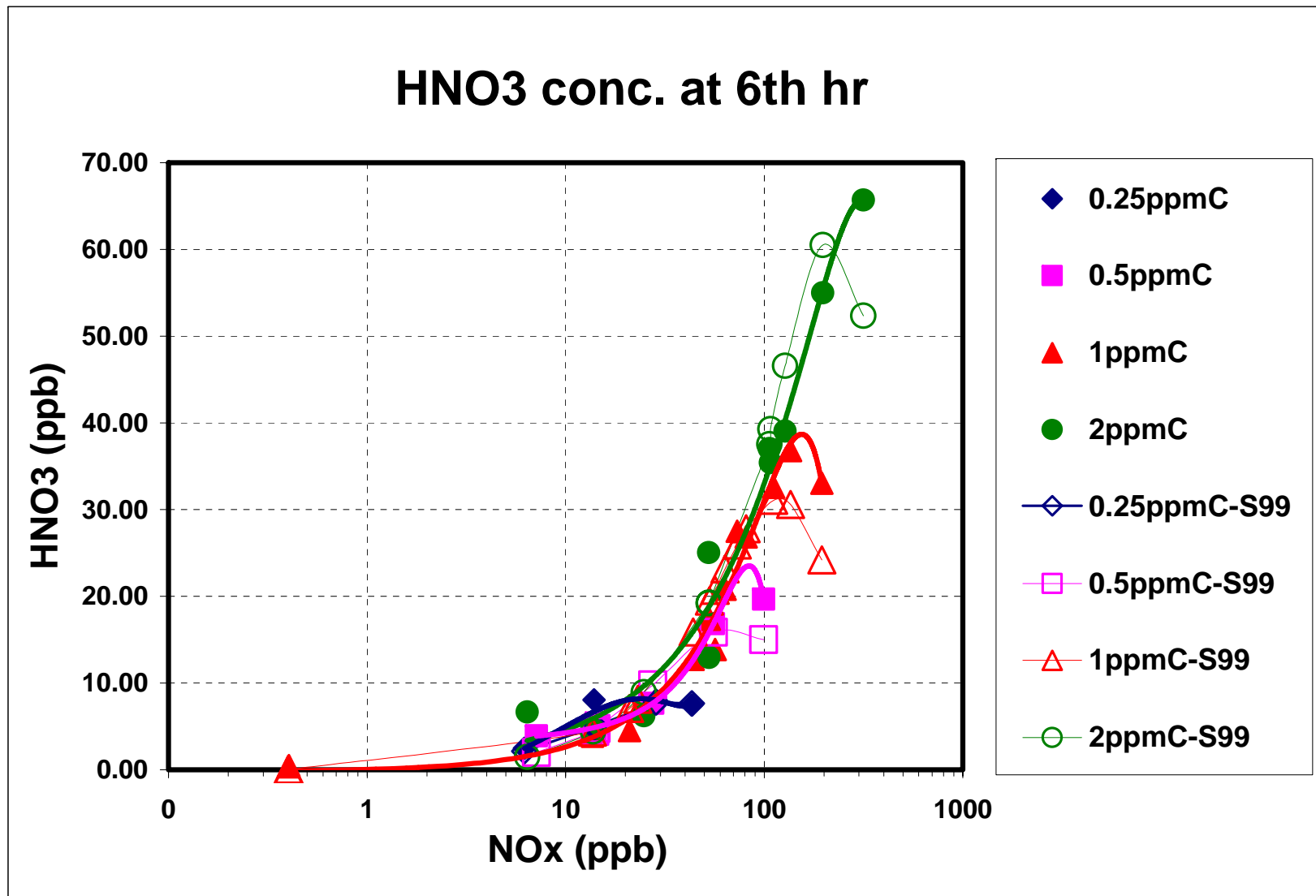


O3 Chamber vs. SAPRC99 Model at 6th hr

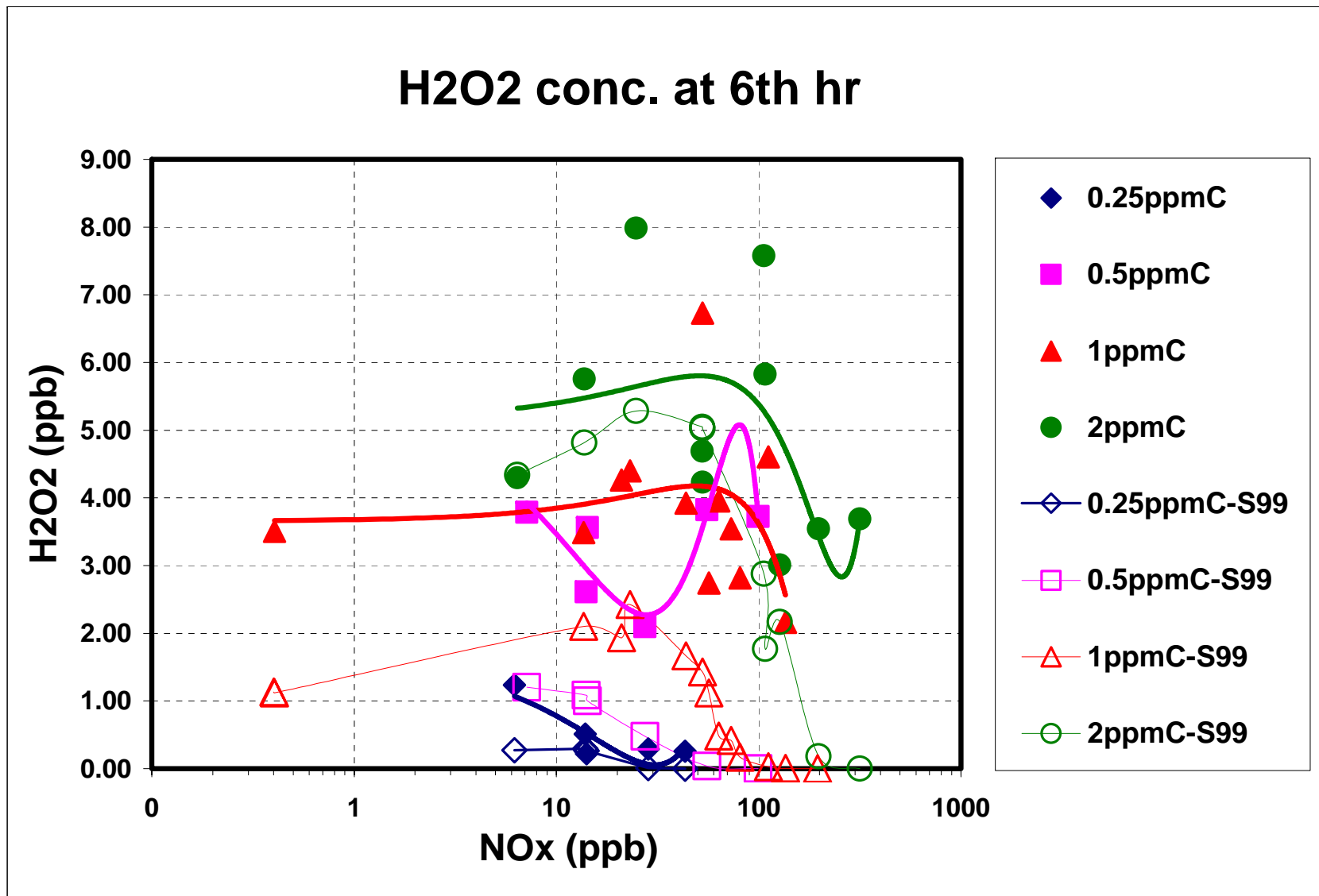


Note: these are not time-series, each data point is the measurement at 360 min for a single chamber experiment.

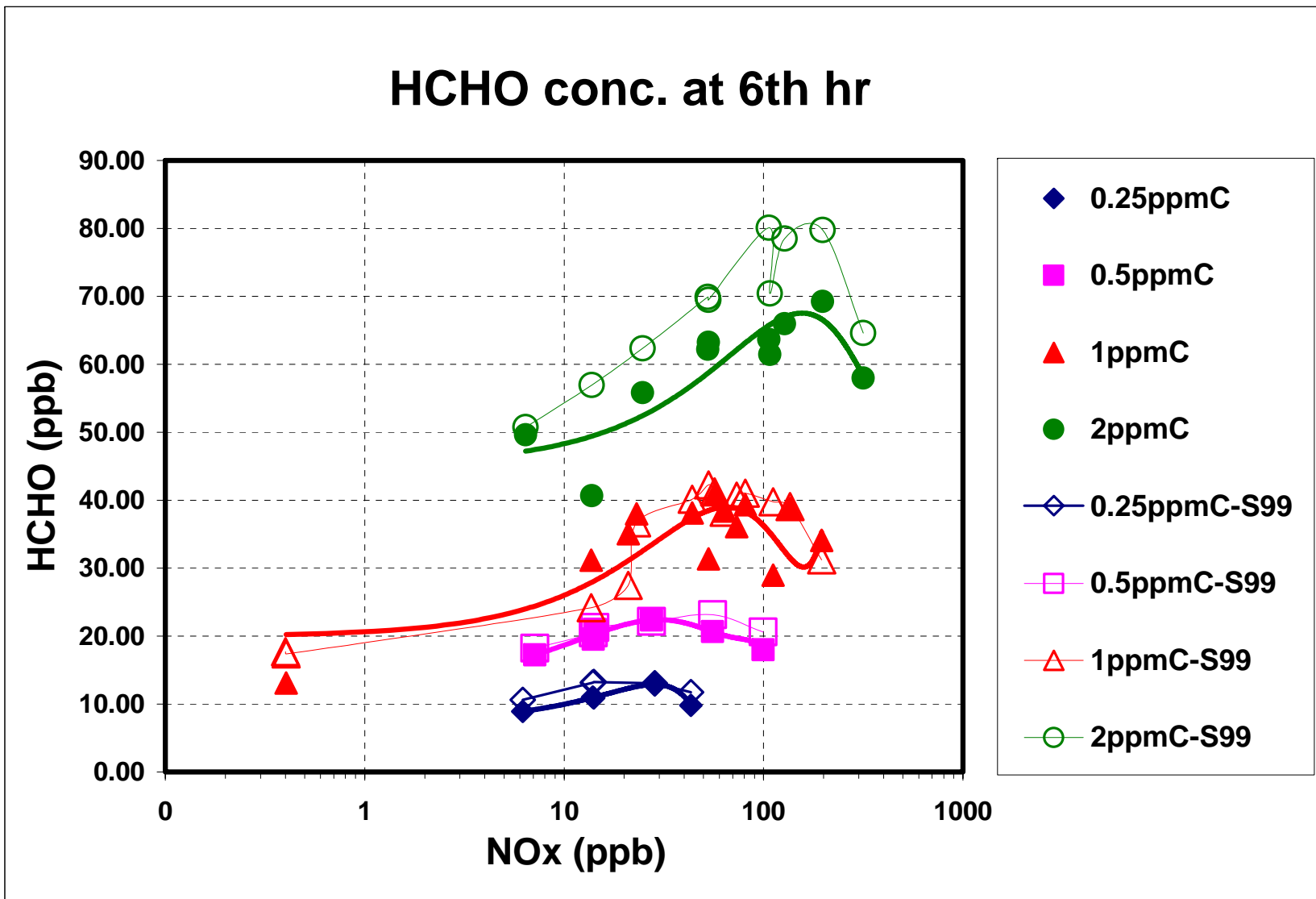
HNO₃ Chamber vs. SAPRC99 Modeled



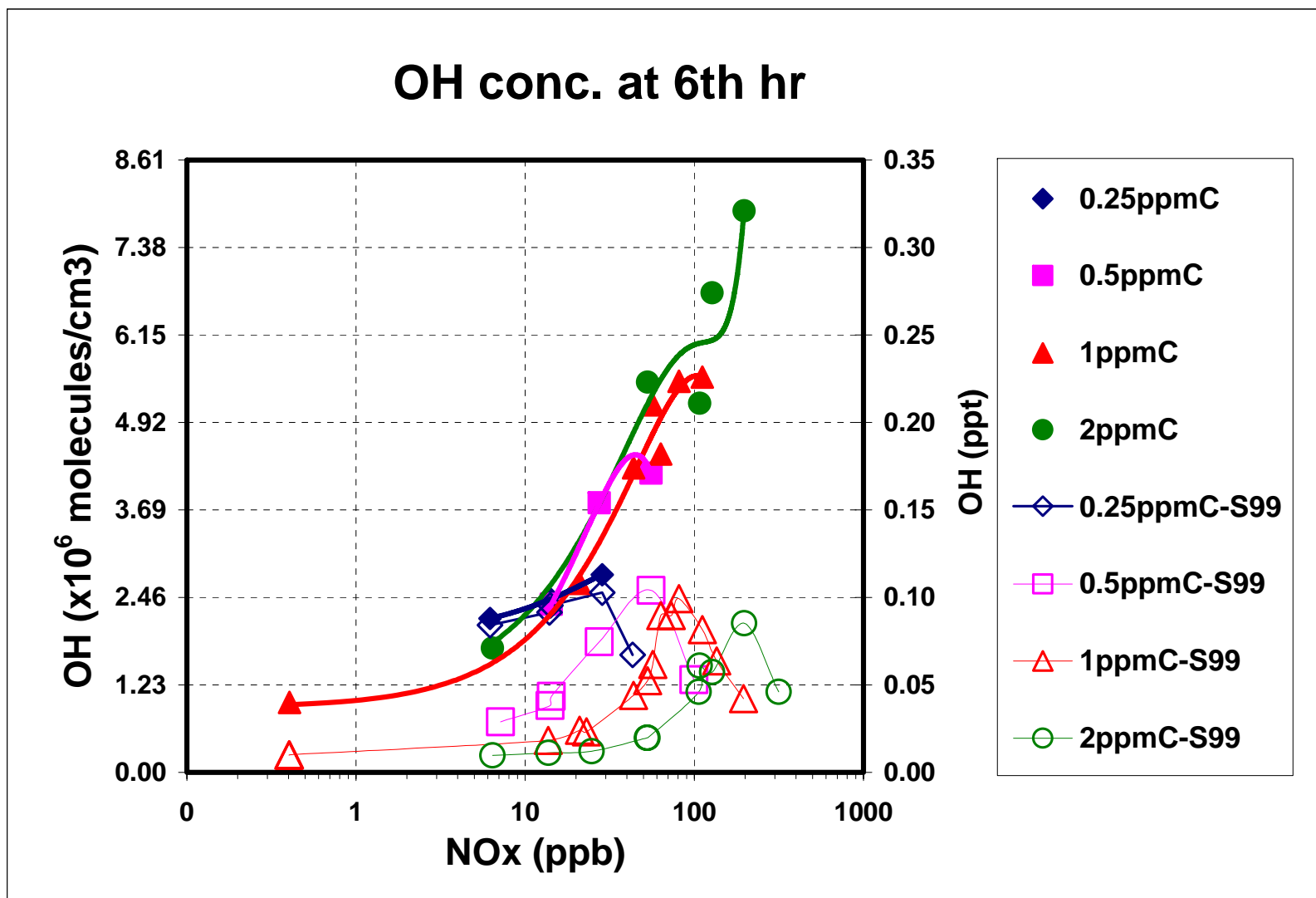
H2O2 Chamber vs. SAPRC99 Modeled



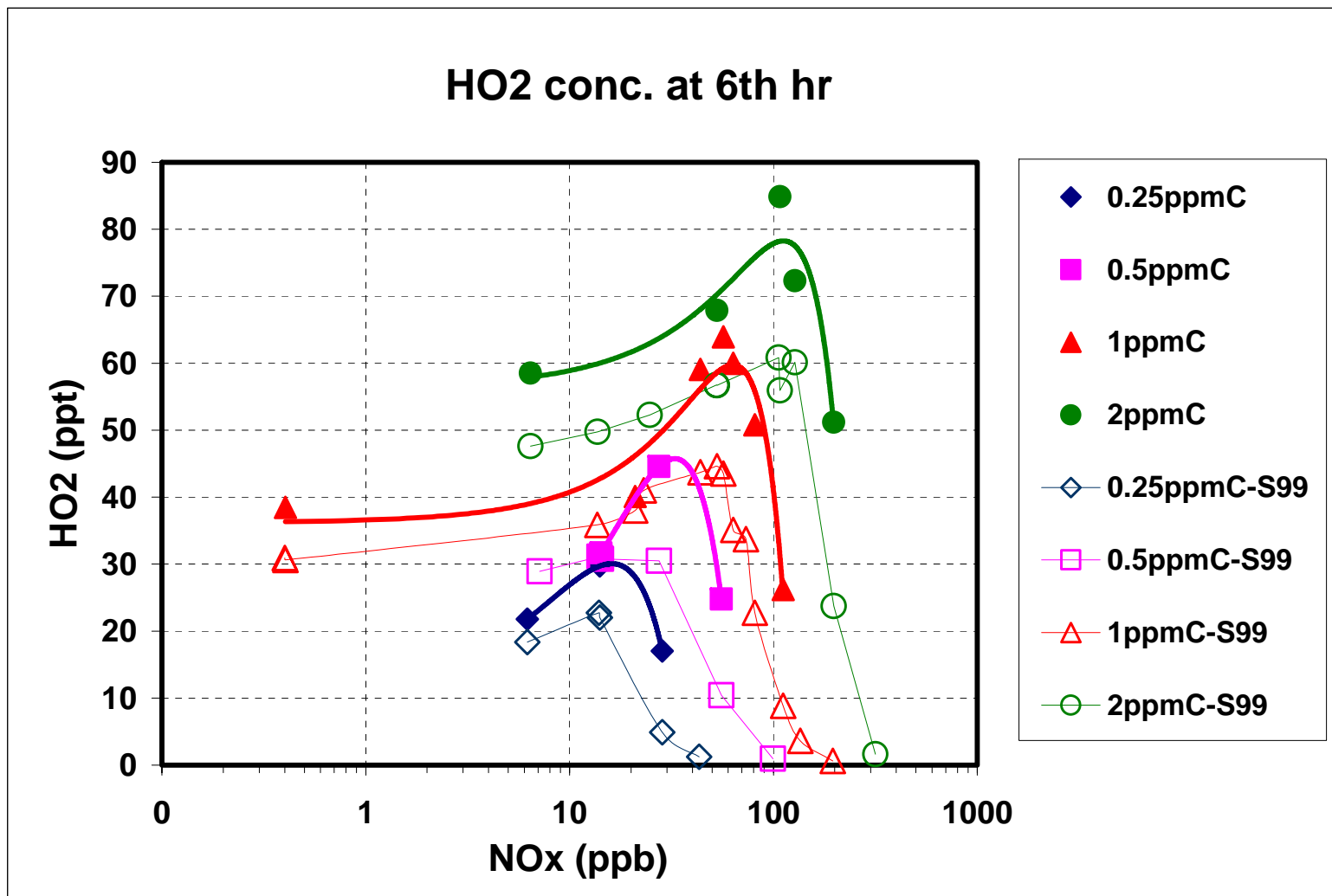
HCHO Chamber vs. SAPRC99 Modeled



OH Chamber vs. SAPRC99 Modeled



HO2 Chamber vs. SAPRC99 Modeled

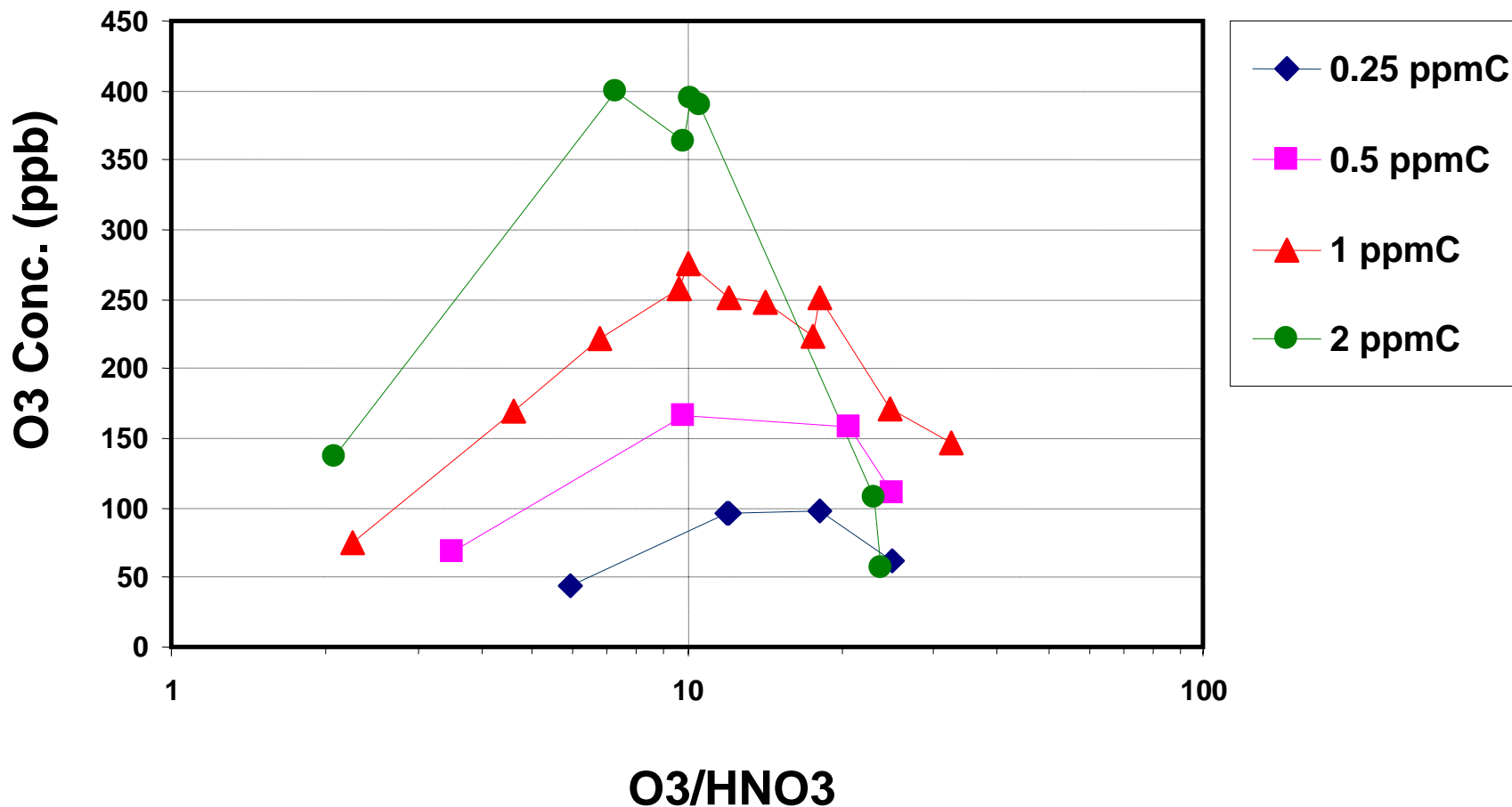


Summary of Model vs. Data

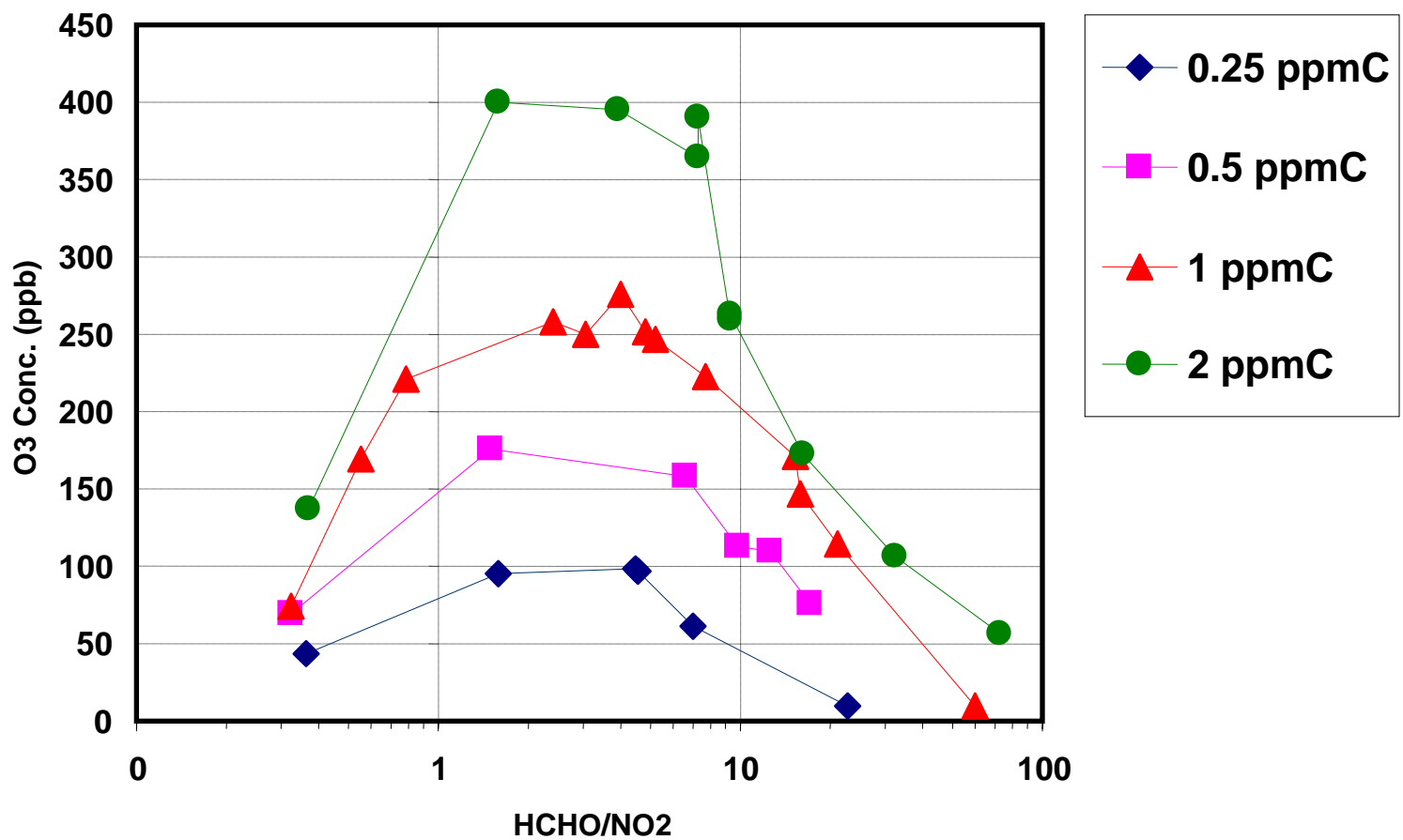
- SAPRC99 has negative O₃ bias at low VOC/NO_x.
- SAPRC99 predicts lower OH and HO₂ vs. LIF.
 - LIF data does not show expected drop in OH at low VOC/NO_x.
- SAPRC99 has large negative bias vs. H₂O₂ data but the H₂O₂ data have large noise.
- SAPRC99 agrees well with data for HNO₃ and HCHO.

O₃ vs. O₃/HNO₃

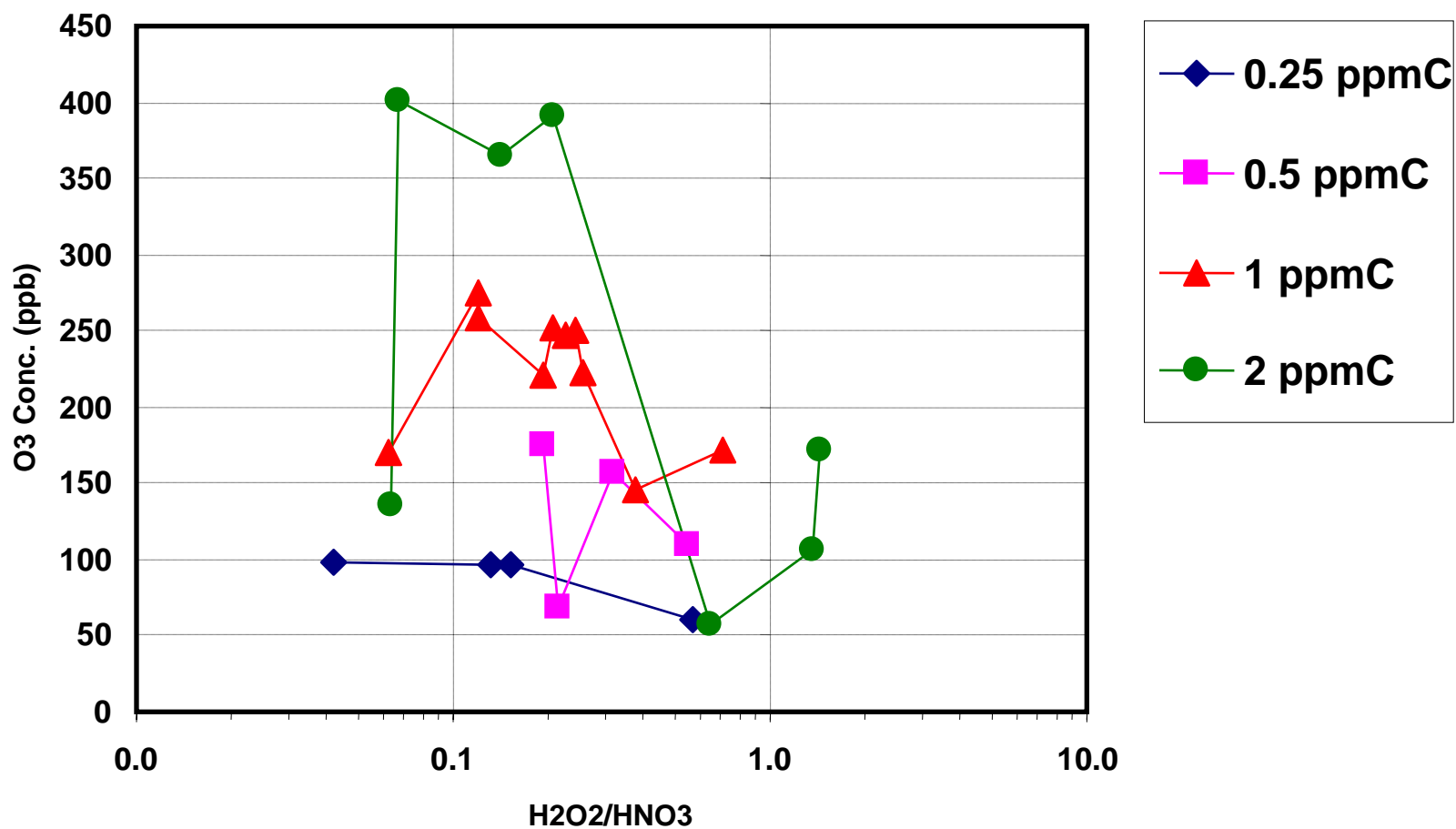
Notes: background O₃ = 0, and no aerosol nitrate formed in chamber.



O3 vs. HCHO/NO2



O3 vs. H2O2/HNO3



Surface Fitting Approach

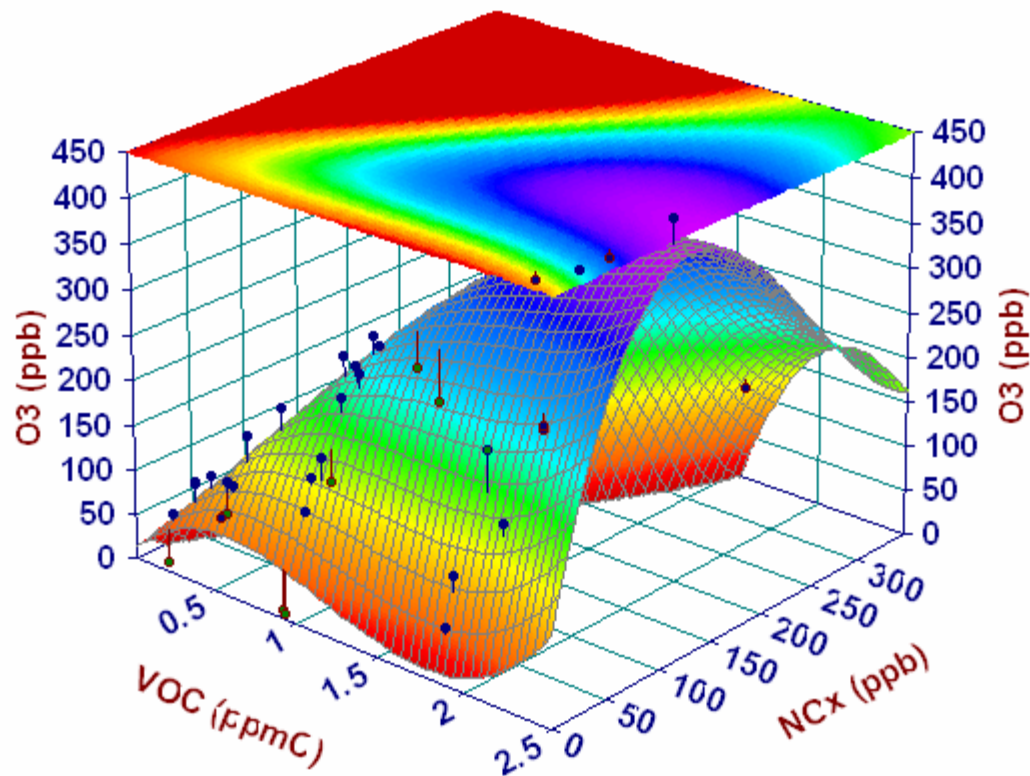
- Initial VOC loading in chamber is not precise:
 - the 1 ppmC experiments had 0.95 to 1.05 ppmC VOC.
 - this shifts the location of experiments on the response surface.
 - makes it difficult to identify ridgeline for level of VOC.
- Solution: Try surface fitting algorithms using all data points to generate a more accurate response surface:
 - TableCurve 3D (Systat Software Inc.) using Chebyshev Bivariate Polynomial equations gave the best fit.

O3 Response Surface

OBM, VOC-NOx-O3 Surface Fit

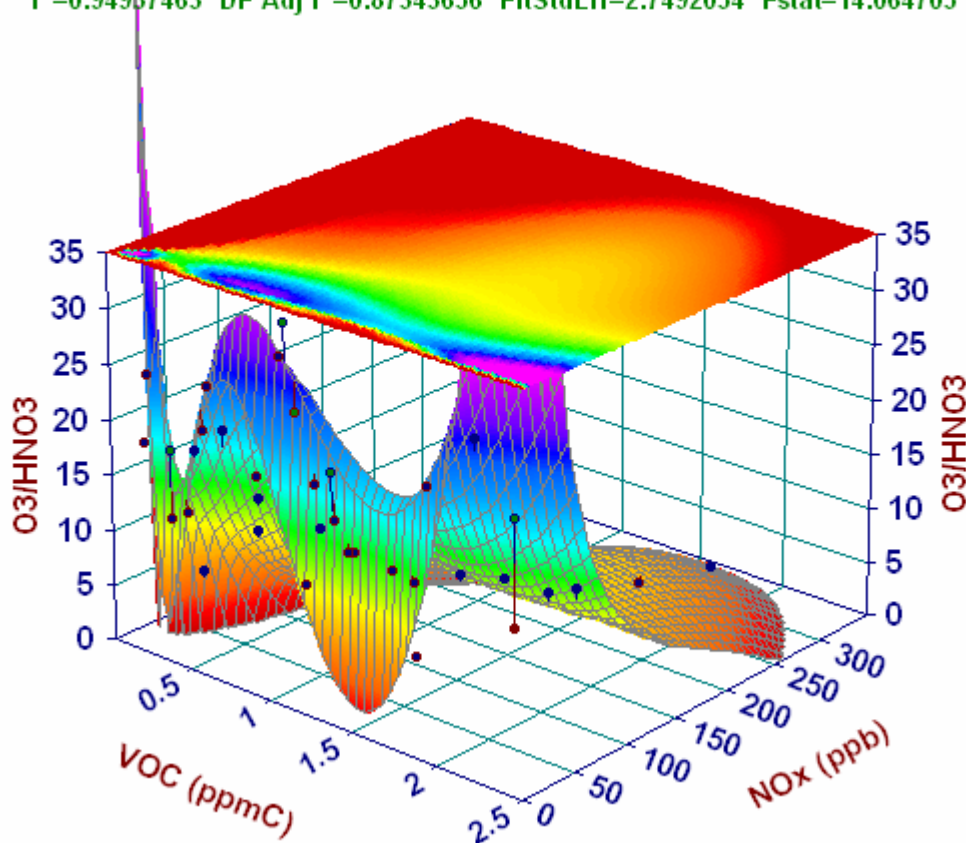
Chebyshev X,Y Bivariate Polynomial Order 3

$r^2=0.94573412$ DF Adj $r^2=0.92402777$ FitStdErr=29.925981 Fstat=50.346939



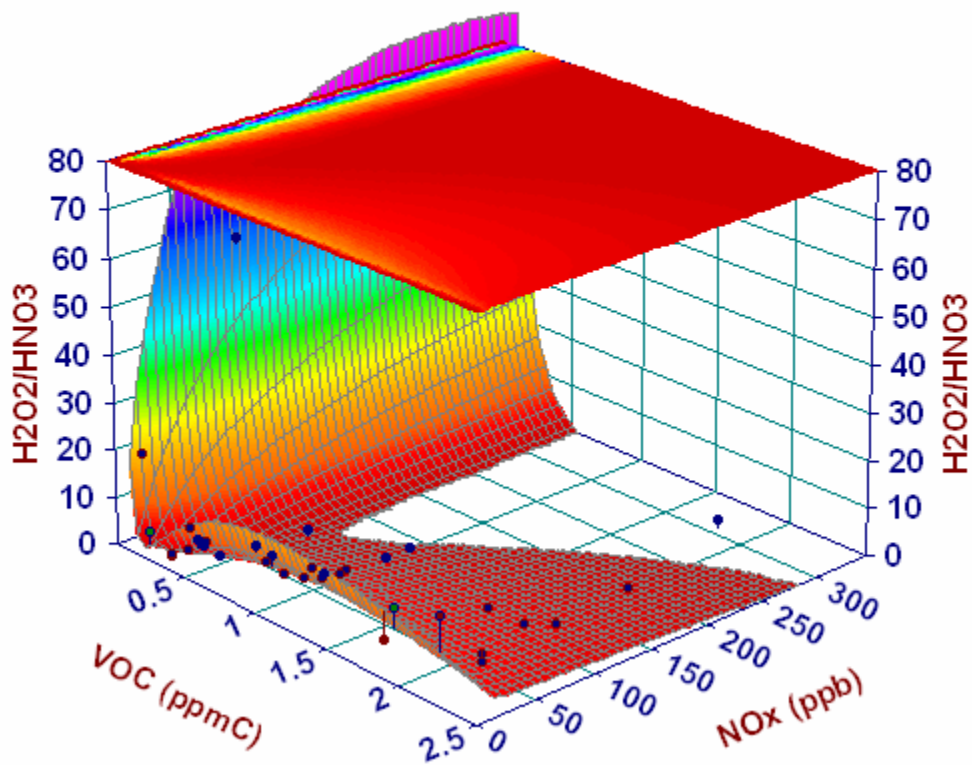
O3/HNO3 Response Surface

OBM. VOC-NOx-O3/HNO3
Chebyshev LnX, LnY Bivariate Polynomial Order 3
 $r^2=0.94987463$ DF Adj $r^2=0.87343656$ FitStdErr=2.7492054 Fstat=14.064705



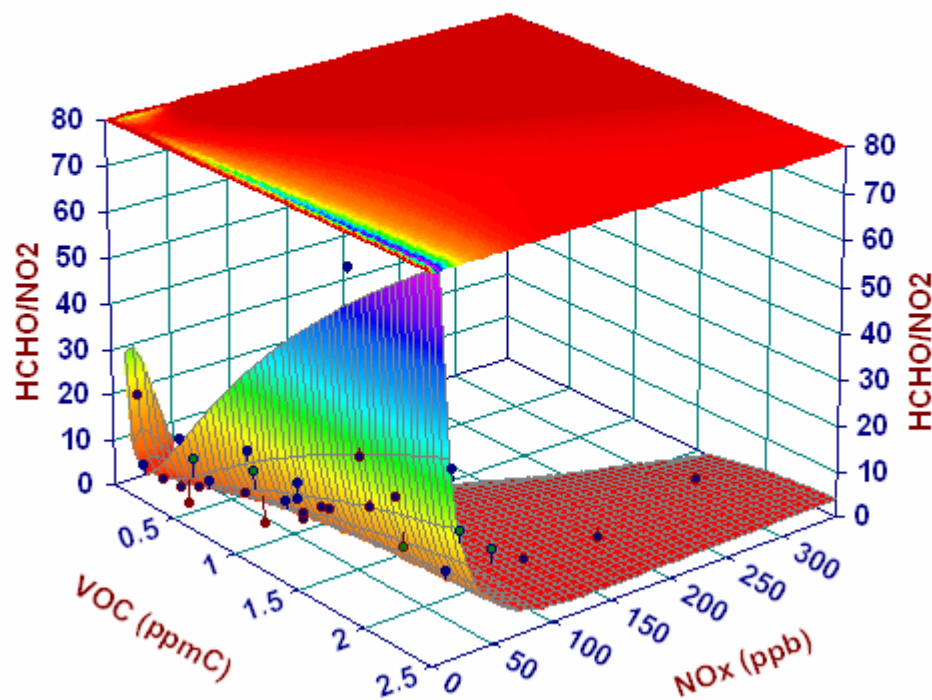
H2O2/HNO3 Response Surface

OBM.VOC-NO_x-H2O2/HNO3
Chebyshev LnX, LnY Bivariate Polynomial Order 3
 $r^2=0.97226897$ DF Adj $r^2=0.96021201$ FitStdErr=2.5437033 Fstat=93.49518



HCHO/NO2 Response Surface

OBM. VOC-NO_x-HCHO/NO₂ Surface Fit
Chebyshev LnX, LnY Bivariate Polynomial Order 3
 $r^2=0.96825292$ DF Adj $r^2=0.95444984$ FitStdErr=2.8093572 Fstat=81.330564

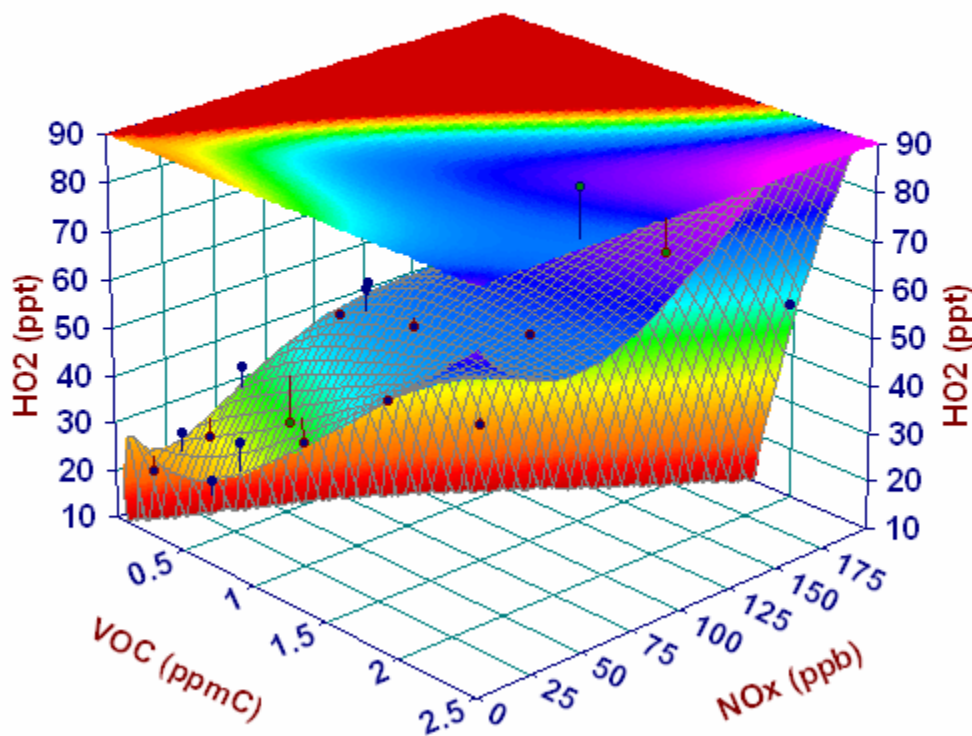


HO2 Response Surface

OBM, VOC-NOx-HO2

Chebyshev LnX,Y Bivariate Polynomial Order 3

$r^2=0.93332719$ DF Adj $r^2=0.83808032$ FitStdErr=7.3003526 Fstat=12.443216

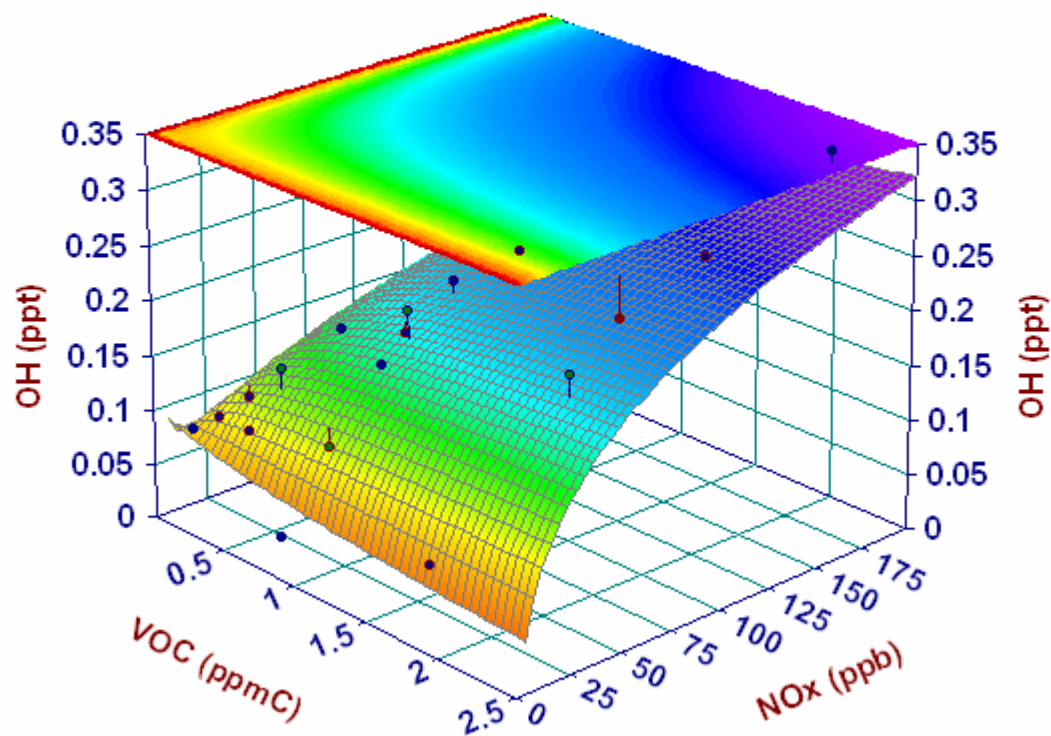


OH Response Surface

OBM.VOC-NOx-OH

Chebyshev LnX, LnY Bivariate Polynomial Order 2

$r^2=0.96022249$ DF Adj $r^2=0.93852566$ FitStdErr=0.017712852 Fstat=57.935599



Transition value of Indicator Ratios Chamber Data vs. Curve Fitting

VOC (ppmC)	O3max (ppb)		O3/NO _x		O3/HNO ₃		H ₂ O ₂ /HNO ₃		HCHO/NO ₂		HO ₂ (ppt)	
	<i>Obs</i>	<i>fit</i>	<i>Obs</i>	<i>fit</i>	<i>Obs</i>	<i>fit</i>	<i>Obs</i>	<i>fit</i>	<i>Obs</i>	<i>fit</i>	<i>Obs</i>	<i>fit</i>
0.25	104	75	5.4	-	11	12.7	0.15	0.15	2.5	3.3	30	7.6
0.5	178	140	4.6	-	15	10.3	0.2	0.09	3	2.2	32	27.5
1	260	264	3.9	-	13	8.3	0.12	0.15	4	3.2	63	45.3
2	420	407	2.9	-	9	9.1	0.10	0.20	4	3.9	90	77.2

Transition value of Indicator Ratios Chamber Data vs. SAPC99 Model

VOC (ppmC)	O3max (ppb)		O3/NOx		O3/HNO3		H2O2/HNO3		HCHO/NO2		HO2 (ppt)	
	<i>Obs</i>	<i>S99</i>	<i>Obs</i>	<i>S99</i>	<i>Obs</i>	<i>S99</i>	<i>Obs</i>	<i>S99</i>	<i>Obs</i>	<i>S99</i>	<i>Obs</i>	<i>S99</i>
0.25	104	94	5.4	7.5	11	18	0.15	0.06	2.5	3.8	30	22
0.5	178	154	4.6	6.8	15	16	0.20	0.10	3	5	32	25
1	260	230	3.9	4.1	13	11	0.12	0.03	4	4	63	38
2	420	380	2.9	4.0	9	9	0.10	0.05	4	7	90	68

Summary of [O₃] Sensitivity Results

- Results generally consistent with previous model sensitivity studies:

Indicator Ratio	Experimental Range	Model Range
O ₃ /HNO ₃ :	8 to 12	about 9
HCHO/NO ₂ :	2 to 4	about 2
H ₂ O ₂ /HNO ₃ :	0.06 to 0.11	0.07 to 0.15

- Small variation in transition ratios as VOC conc changes for O₃/HNO₃ and for H₂O₂/HNO₃.

Caveats in Interpretation of Results

- Cannot exactly reproduce VOC in each experiment:
 - “constant” VOC series can vary by +/-10%
- Problem with high noise in TDL H_2O_2 measurements.
- No background O_3 or H_2O in chamber, this might bias results relative to ambient air.

Conclusions

- O_3/HNO_3 , O_3/NO_x and $HCHO/NO_2$ seem useful
 - O_3/NO_x data is most widely available but “ NO_x box” measurements are too uncertain.
 - in ambient air we need to measure the sum $HNO_3+NO_3^-$
- H_2O_2 is too difficult to measure.
- Indicators are not robust in chamber nor in model.
 - value of indicator associated with O_3 ridgeline varies depending on VOC level.

Use of Indicators

- If they work indicators predict whether O₃ is more sensitive to VOC or NO_x at a particular set of conditions
 - Indicators do not predict how much emissions control is needed to attain air quality goals.
 - Indicators do not determine if attainment of NAAQS is feasible with a VOC or NO_x control strategy.
- Indicators should be used to gain more confidence in the model simulations that are used to develop emissions controls.

Caveat for Application in ambient air

- Chamber has low water vapor therefore low OH production from $\text{H}_2\text{O} + \text{O}^1\text{D}$ therefore it is possible that performance of indicators differs in chamber versus ambient air.
- To account for effect of O^1D on indicators, we need to develop improved chemical mechanism and use model simulations of ambient air with improved mechanism to further evaluate OBMs.